

Atomistic understanding of hundred thousand tons: the new steel story

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Self-organized nanostructuring by selective phase transformation

Motivation for nanostructuring alloys via self-organization:

Properties of alloys change profoundly at nanoscale (mech. properties)

**Self organization essential for nanostructuring large quantities: e.g.
steels: 1.4 Billion t/a; aluminum: 24 Million t/a; titanium: 150 Thousand t/a**

Can be applied to many metallic alloy systems that are the backbone of modern societies



Design alloys



Solidification



Processing



Great products



Knowledge-based
(ab-initio, DFT, GGA)



'Rapid' solidification
(of 100 tons)



TMP - induced
nanostructures



Design complexity

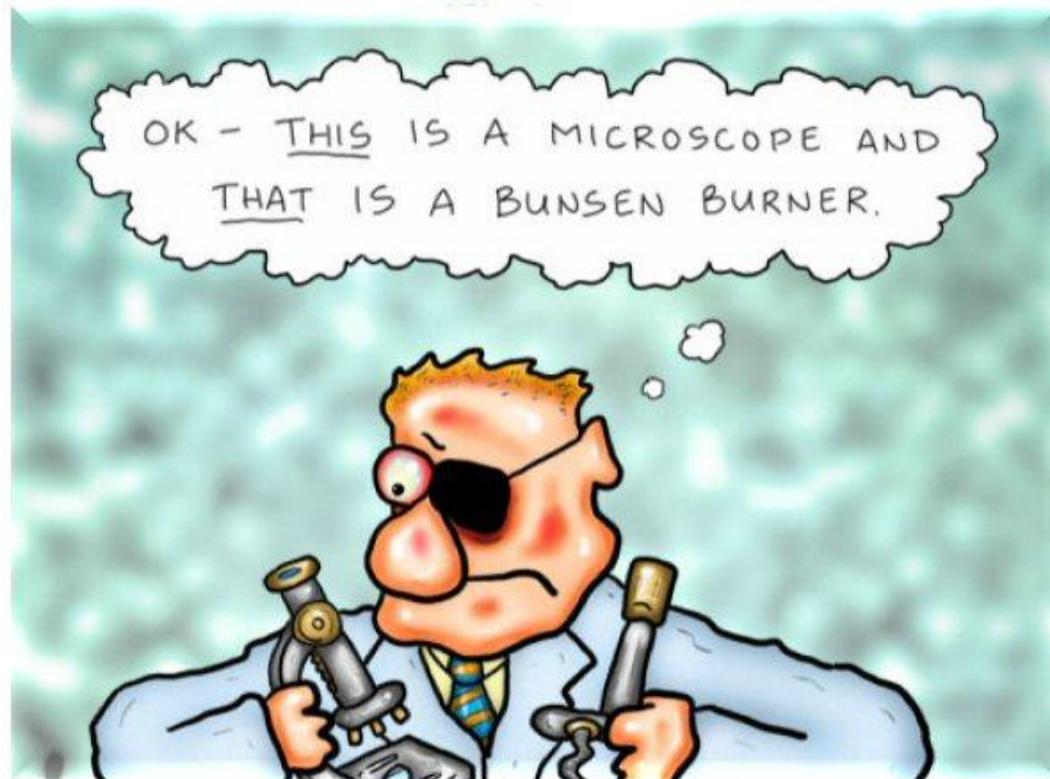


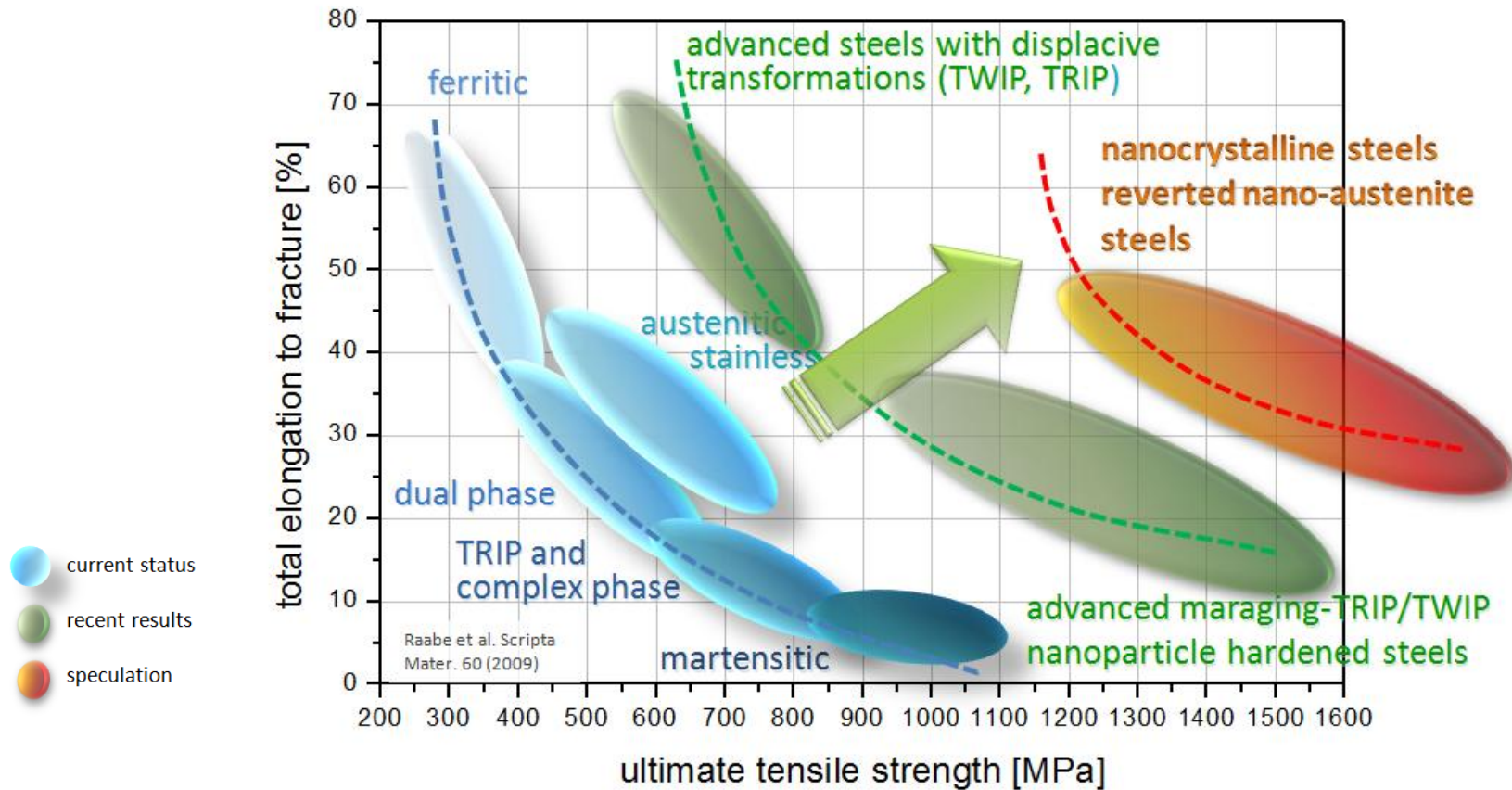
Get the right tools for the job:

Models, characterization, new mechanisms, synthesis

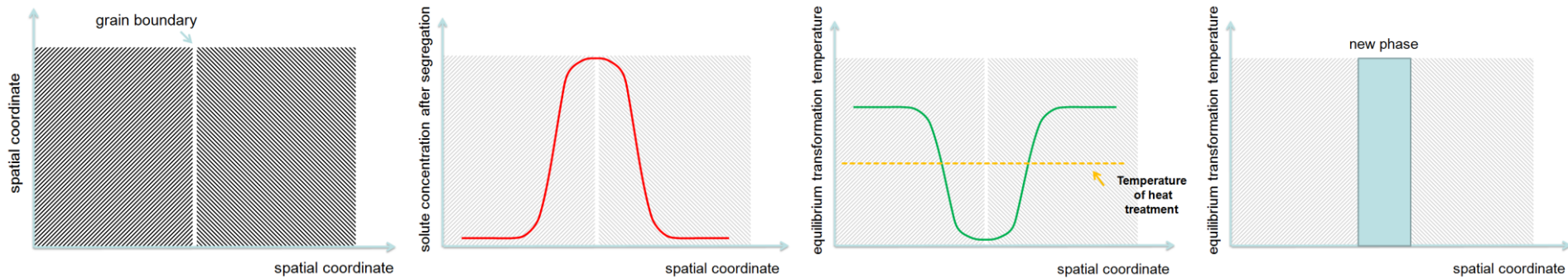
things
that don't
work

things that work





Inverse strength-ductility relation (strain hardening where needed; use smart mechanisms) ?



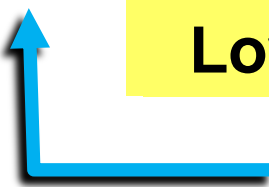
Driving force for solute decoration

- Gibbs adsorption isotherm: equilibrium segregation
- Non-equilibrium segregation (additional point defects)
- Kinetic freezing

Hetero-Interfaces Homophase-Interfaces



Steel	C	Ni	Co	Mo	Ti	Al	Mn	Fe
Maraging	0.01	18	12	4	1.6	0.15	0.05	Balance
09MnPH	0.01	2	-	1	1.0	0.15	9	Balance
12MnPH	0.01	2	-	1	1.0	0.15	12	Balance
15MnPH	0.01	2	-	1	1.0	0.15	15	Balance



Low carbon: ductile martensite

Precipitation Hardenable

Mn (+Ni): austenite (TRIP)

Vacuum induction melting

Annealing

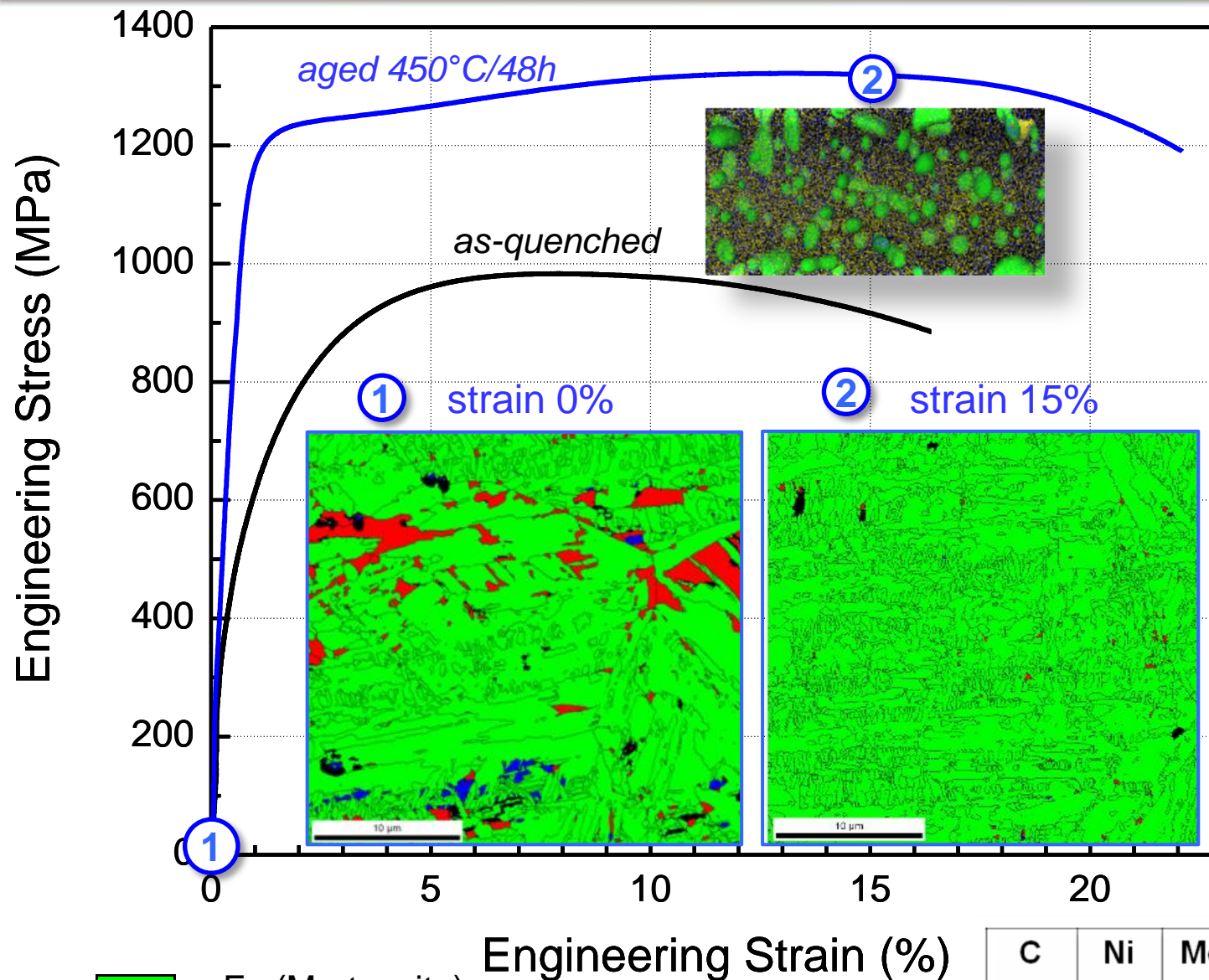
Hot deformation

Solution heat treatment

Quenching \Rightarrow **Martensite** + retained austenite

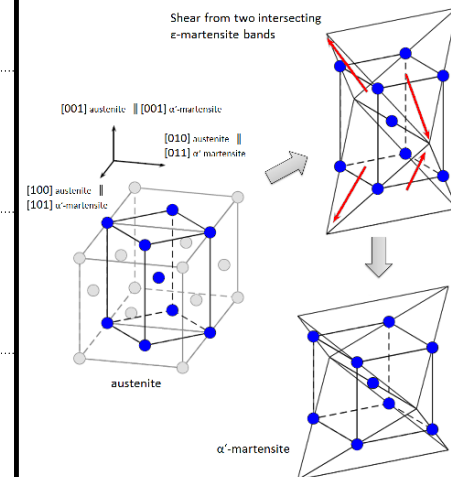
Aging (450°C) “**Maraging**” retained + new austenite

Effect of aging on ductility



Precipitation
hardening

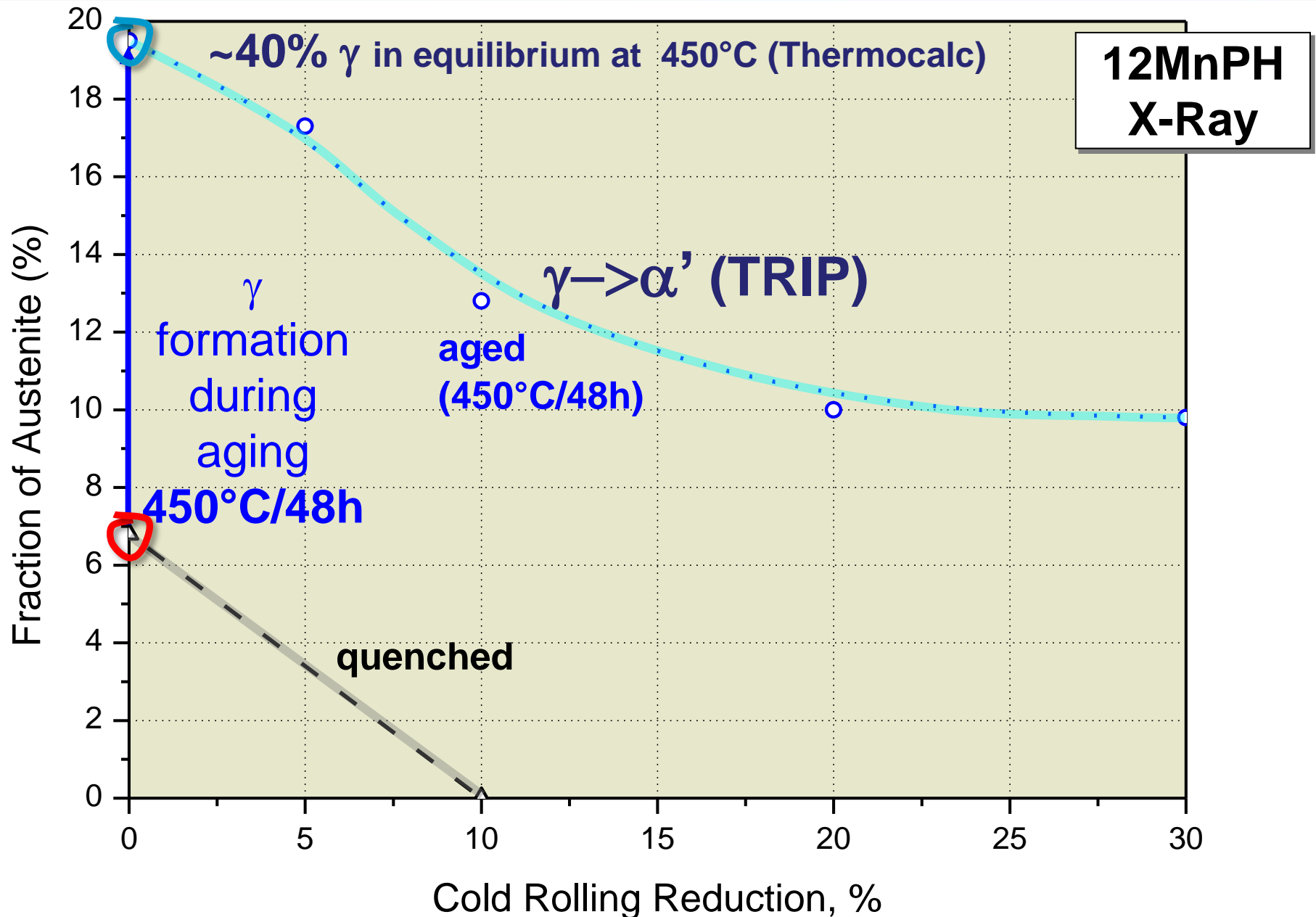
increase of austenite
fraction during aging

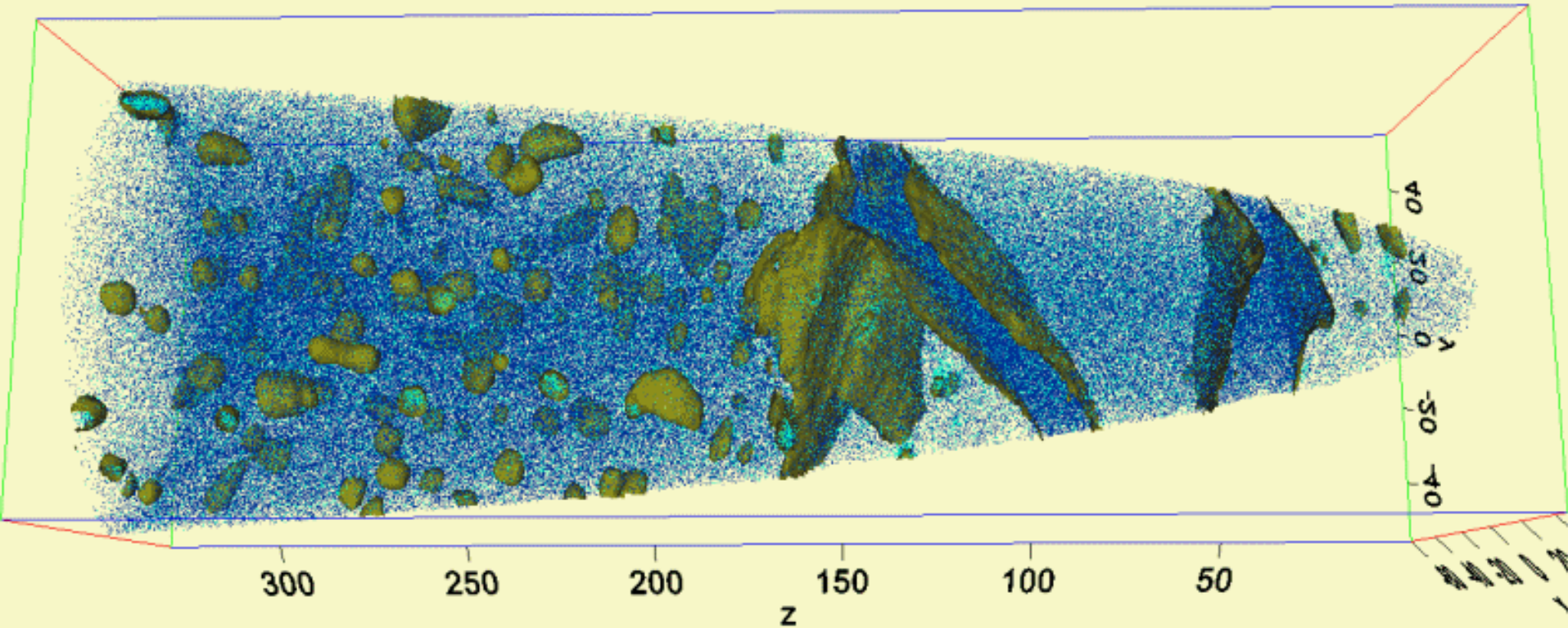


■ α -Fe (Martensite)

■ γ -Fe (Austenite), vol. fraction 15-20%

C	Ni	Mo	Ti	Al	Mn	Fe
0.01	2.0	1.0	1.0	0.15	12	bal.





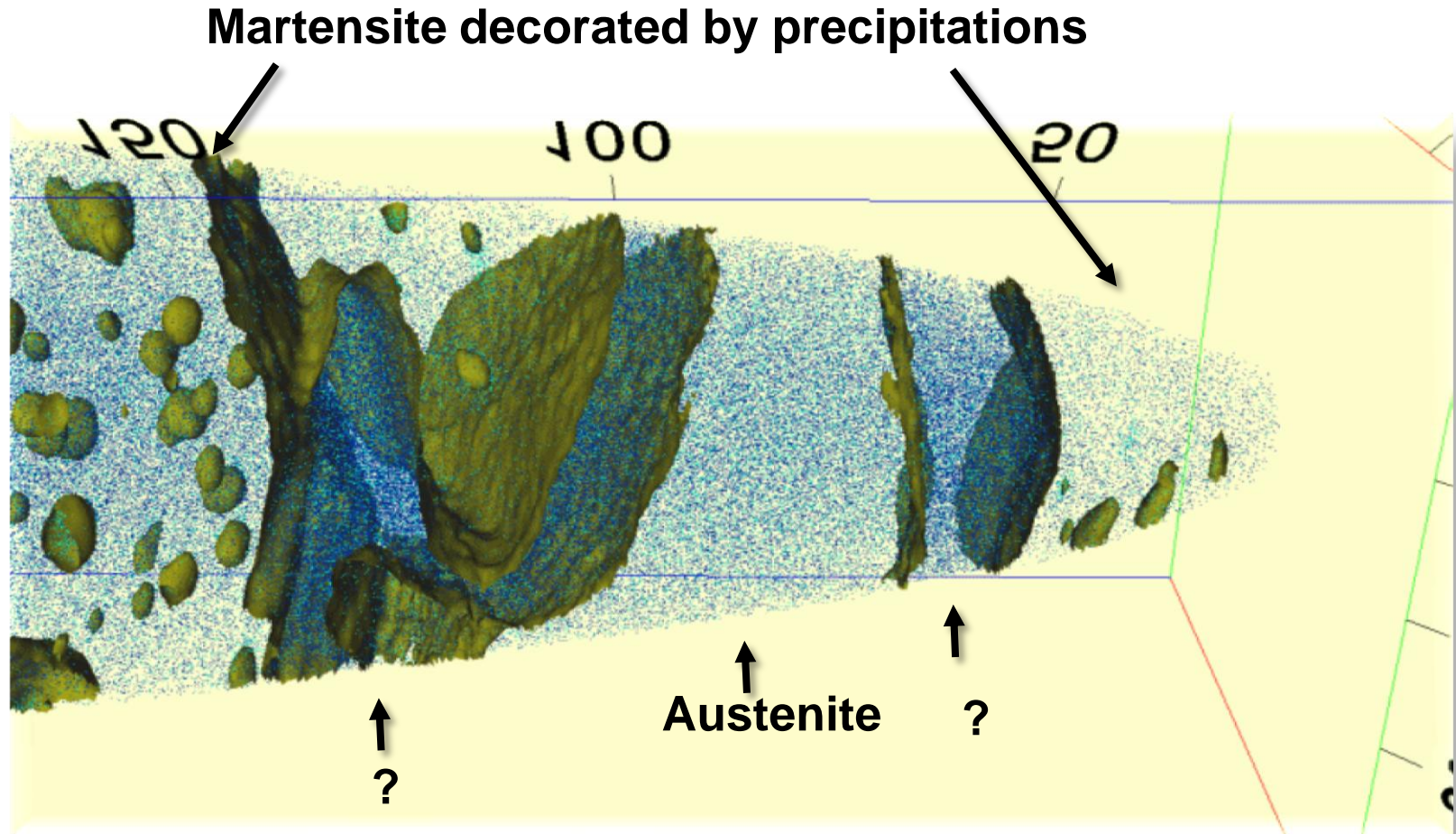
Mn atoms

Ni atoms

Mn iso-concentration: 18 at.%

C	Ni	Mo	Ti	Al	Mn	Fe
0.01	2.0	1.0	1.0	0.15	12	bal.

70 million ions
Laser mode
(0.4nJ, 54K)



Mn atoms

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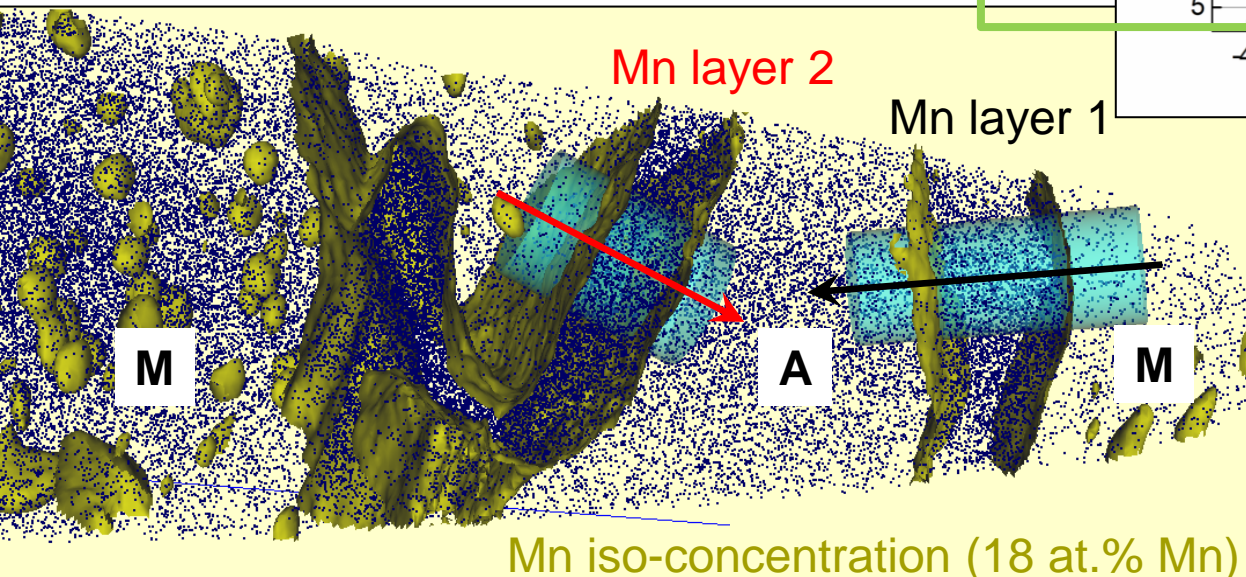
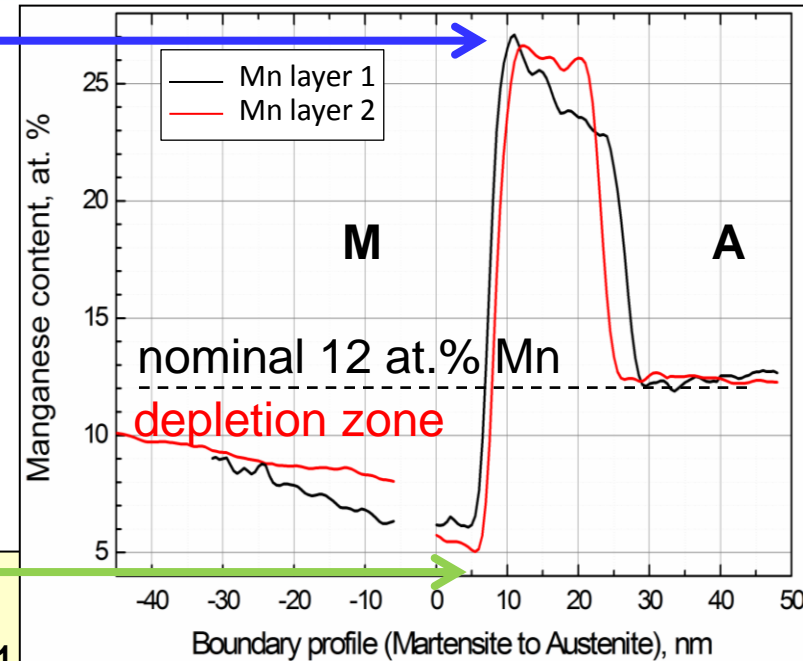
70 million ions
Laser mode
(0.4nJ, 54K)

Thermo-Calc \Rightarrow

equilibrium Mn-conc.:

27 at. % Mn in austenite (A)

3 at. % Mn in ferrite (martensite) (M)



precipitates in α'

$$x_{Diff} \cong 2\sqrt{Dt} \cong 30nm$$

no precipitates in
austenite

$$x_{Diff} \cong 2nm$$

men [STEM BF]
TEM 200kV x400k 50%

200.0nm

Thermo-Calc \Rightarrow

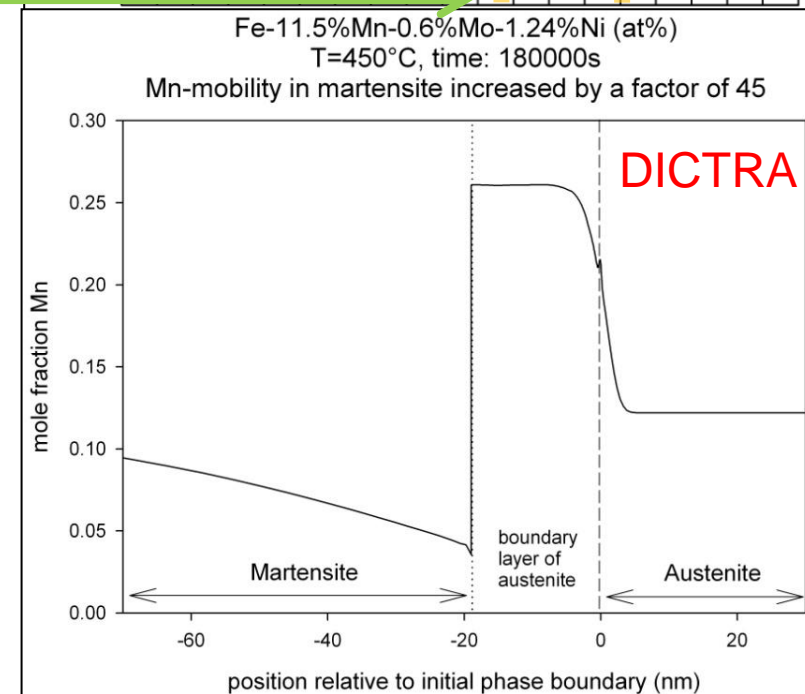
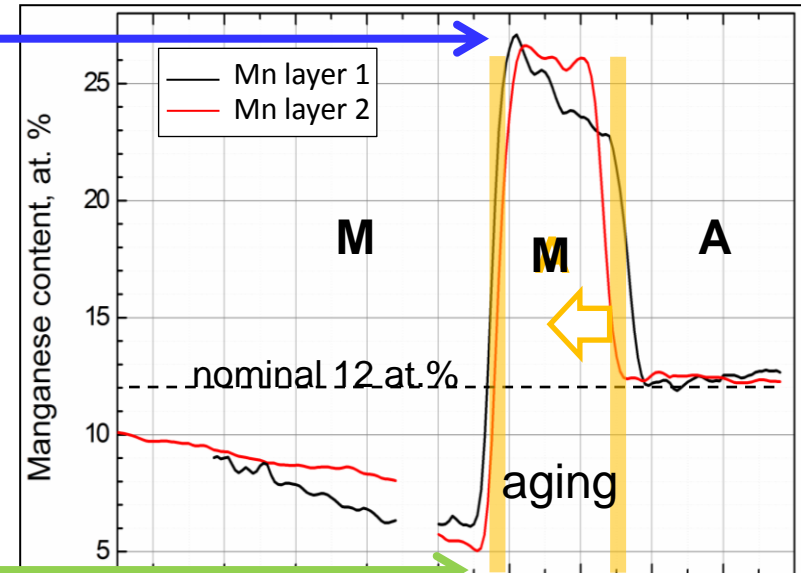
equilibrium Mn-conc.:

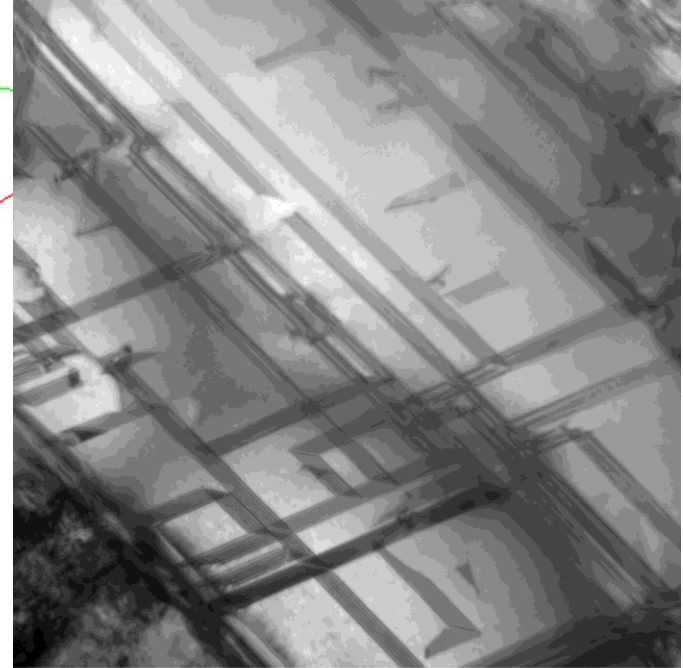
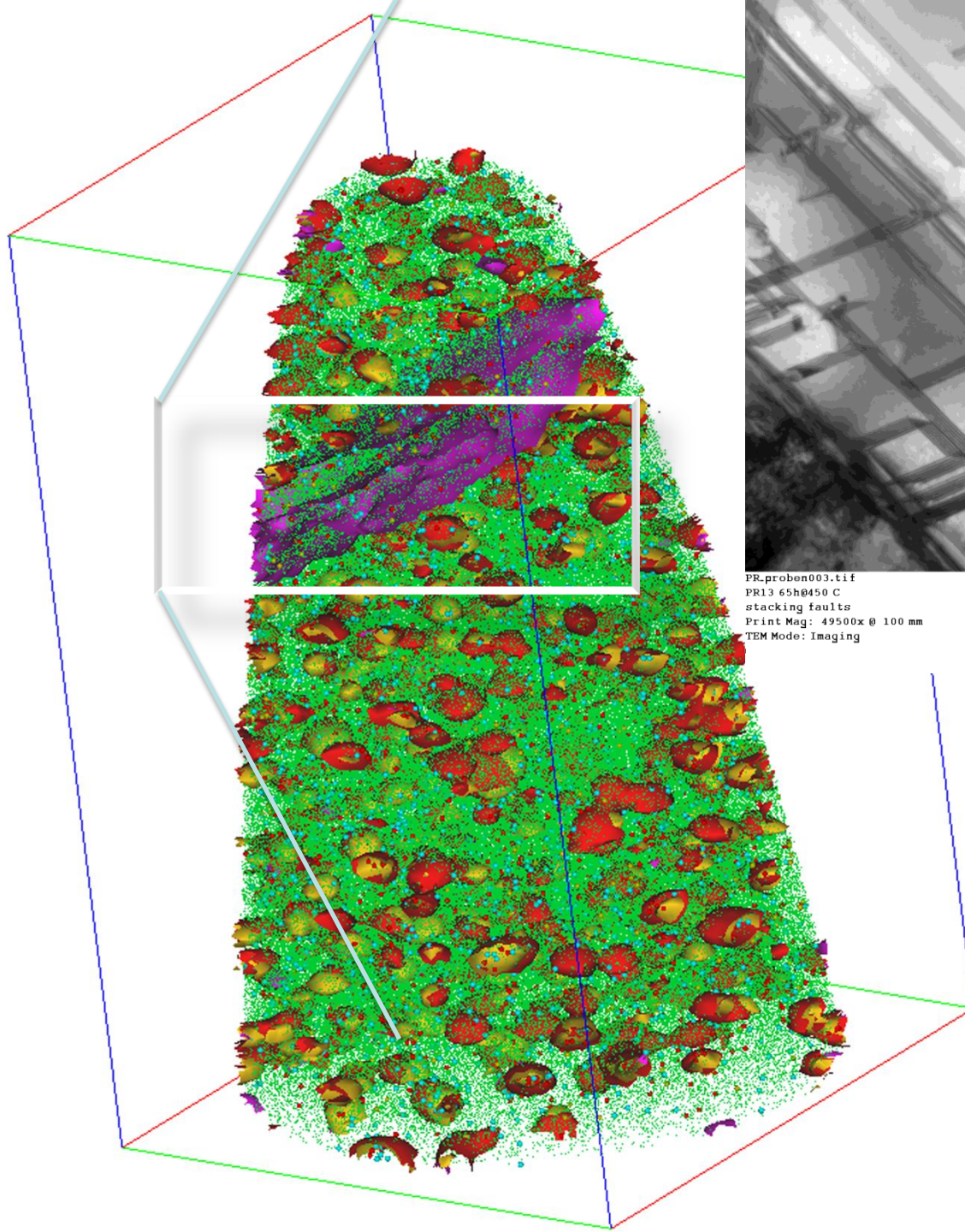
27 at. % Mn in austenite (A)

3 at. % Mn in ferrite (martensite) (M)

**Excellent agreement between
experiment & simulation !**

**Kinetic freezing and
associated austenite reversion !**





PR.proben003.tif
PR13 65h@450 C
stacking faults
Print Mag: 49500x @ 100 nm
TEM Mode: Imaging

100 nm
HV=200,0kV
Direct Mag: 44000x
X:-107.789 Y: 194.272
MPI f r Eisenforschung - CM20

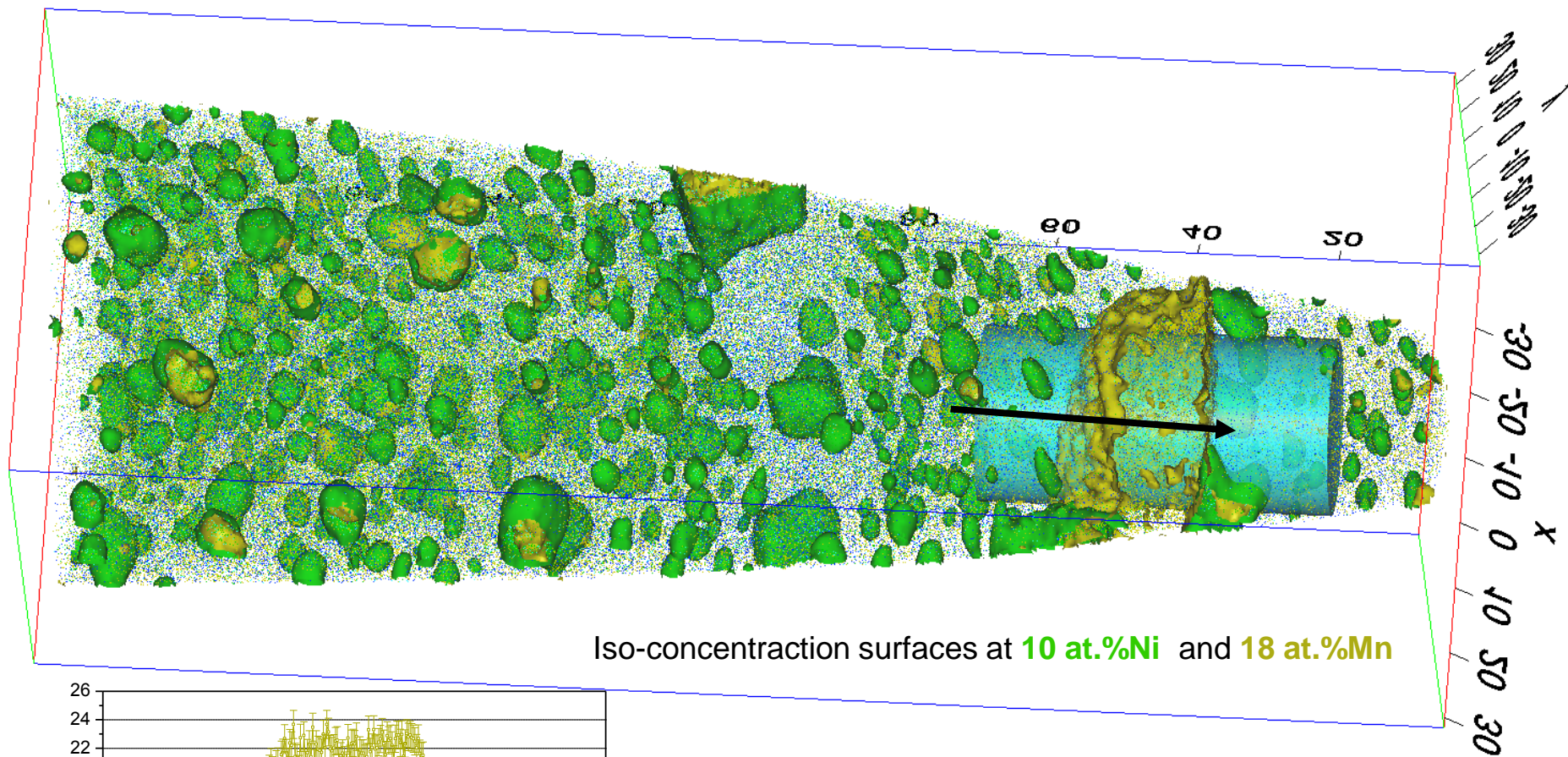
● Ni ● Mn
● Al ● Fe

■ Mn surface (16at.%)

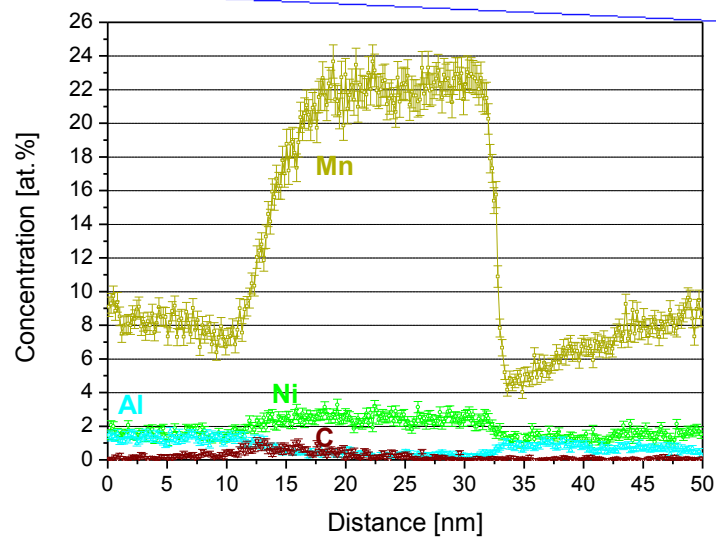
■ Ni surface (6 at.%)

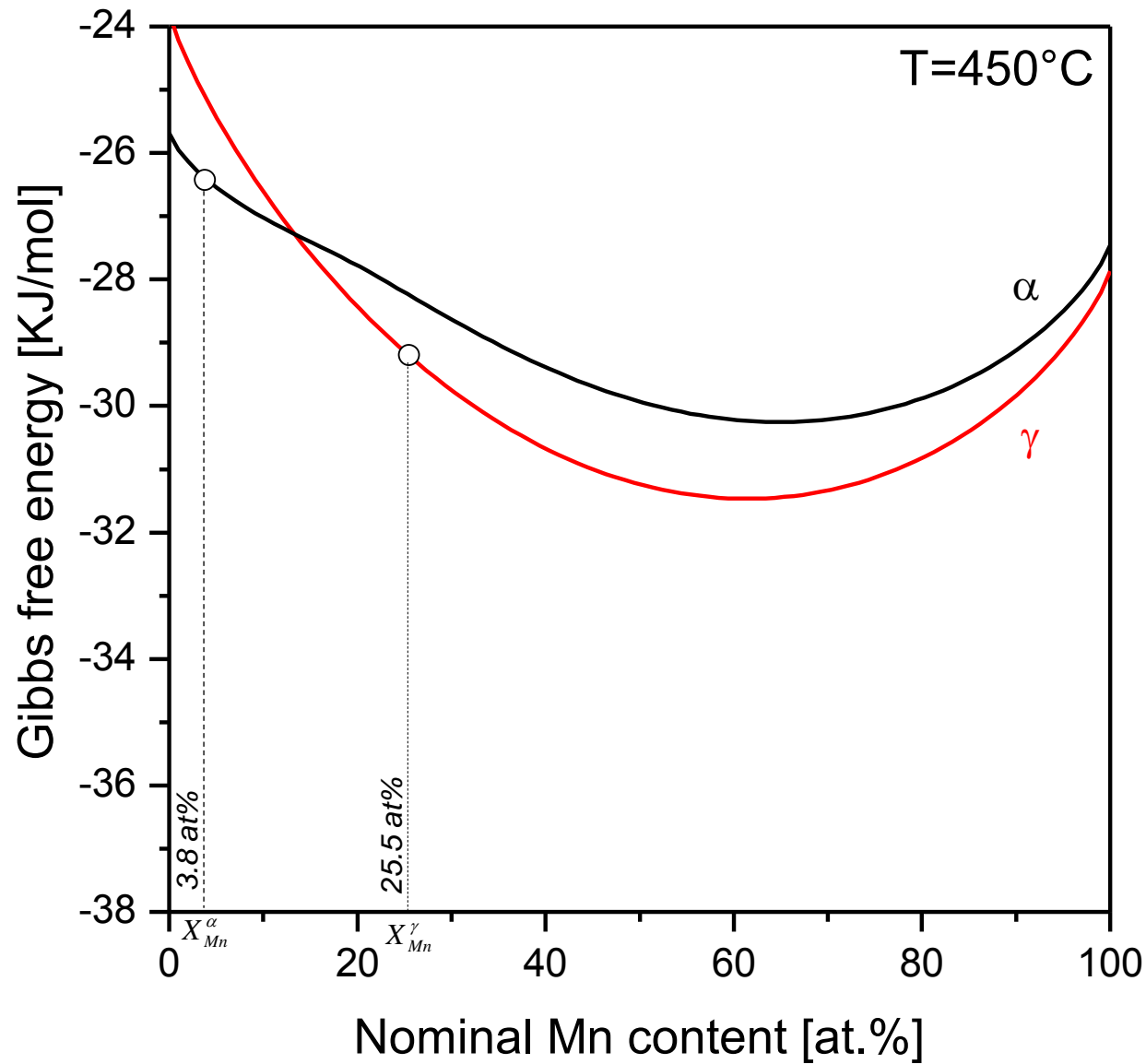
■ Al surface (6 at.%)

10 nm



Iso-concentration surfaces at **10 at.%Ni** and **18 at.%Mn**





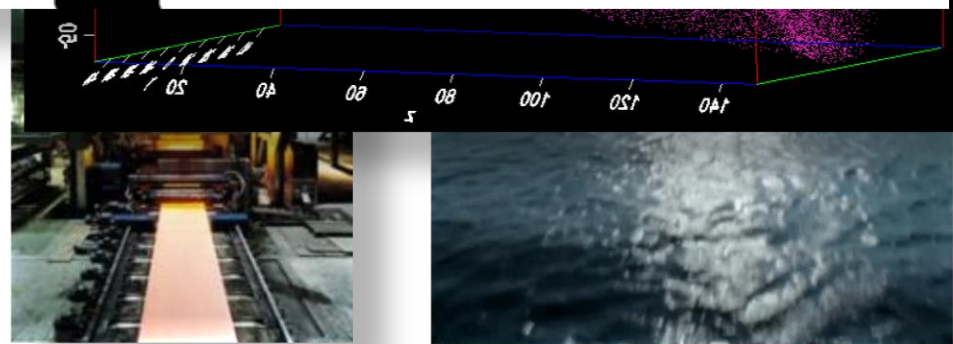
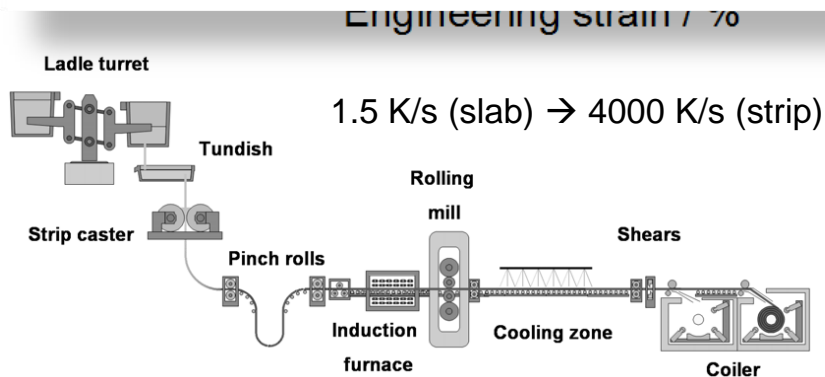
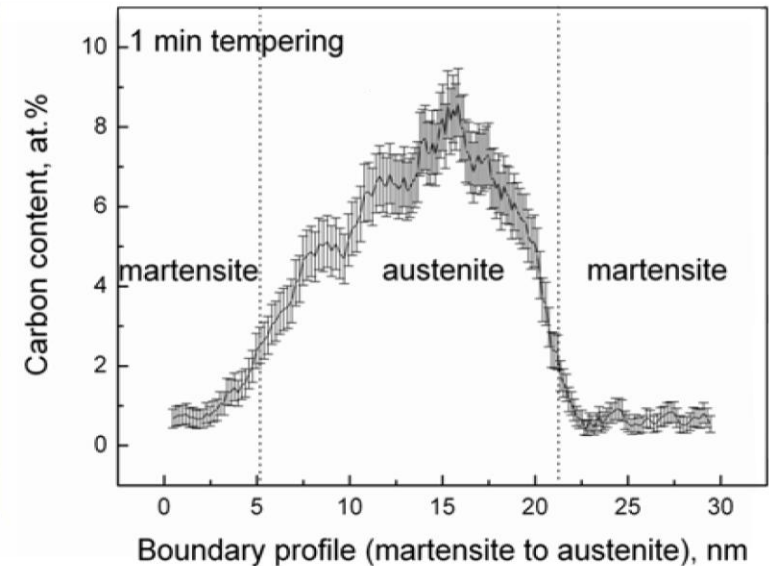
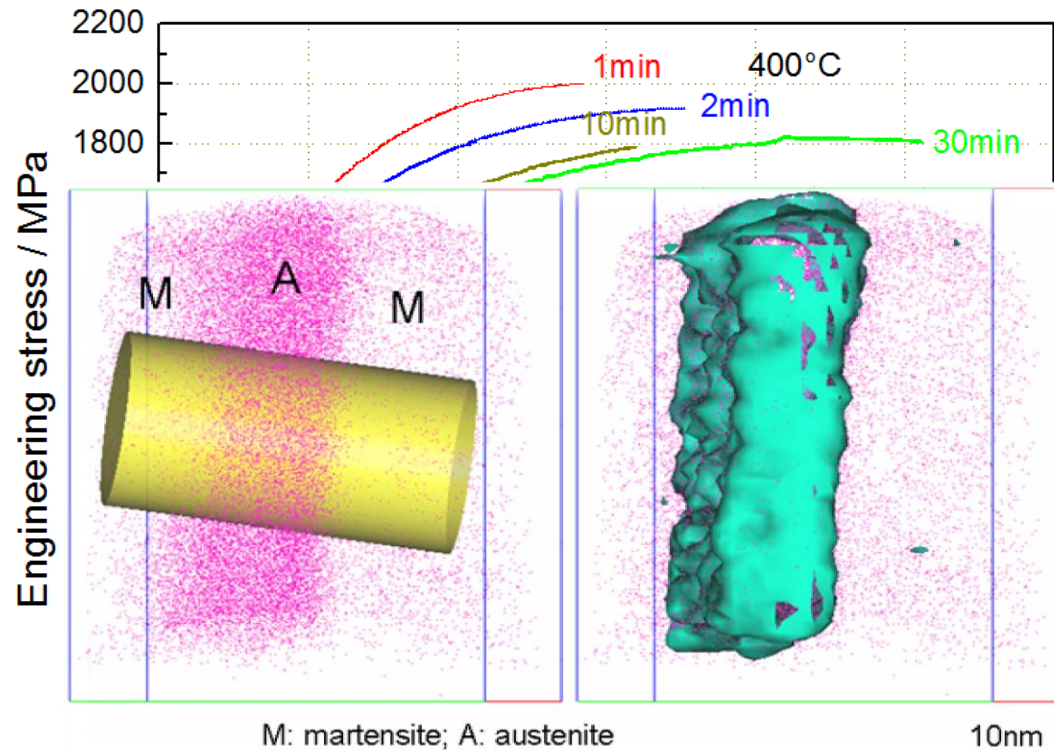
Martensite relaxation & aging & nanoscale austenite reversion



from 650 MPa to 2 GPa (SAME alloy: Fe 13 Cr > 0.3 C)

Aged martensite: Ms-relaxation + prec. (aging) + austenite reversion

Ultra high strength and corrosion resistance

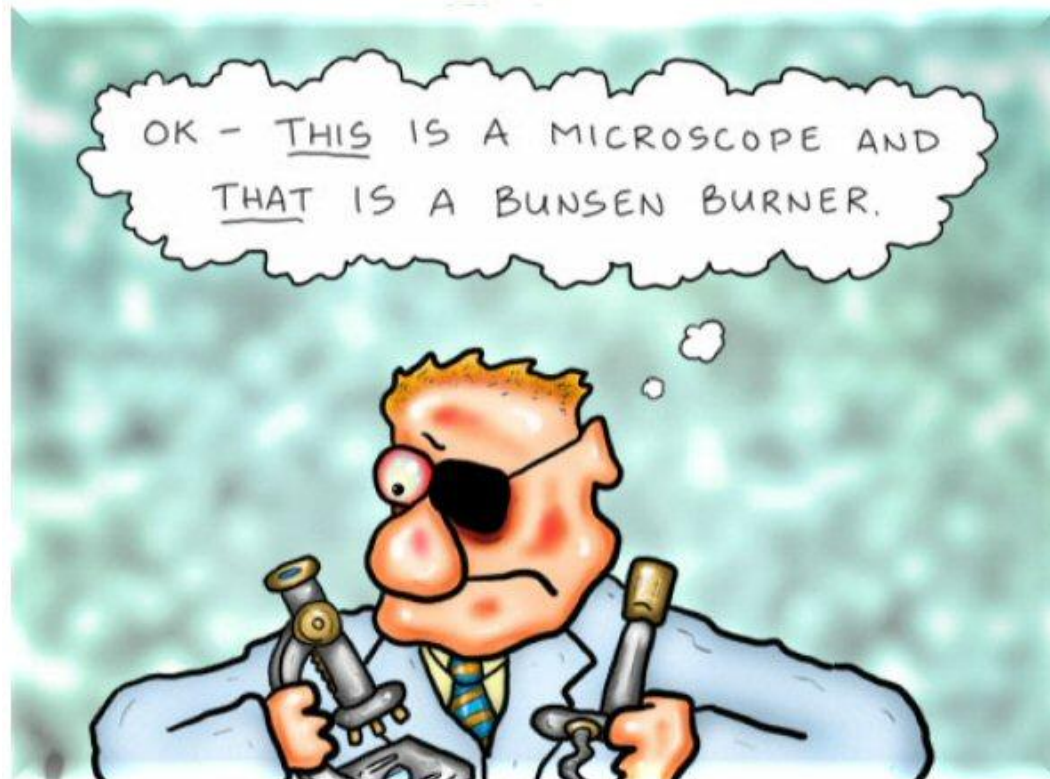


Get the right tools for the job:

Models, characterization, new mechanisms, **synthesis**

things
that don't
work

things that work



Challenge: How to obtain large numbers of mechanical data sets for new structural metallic alloys:

systematic and fast variations in:

composition

TM processing

Heat treatment

Mechanical testing

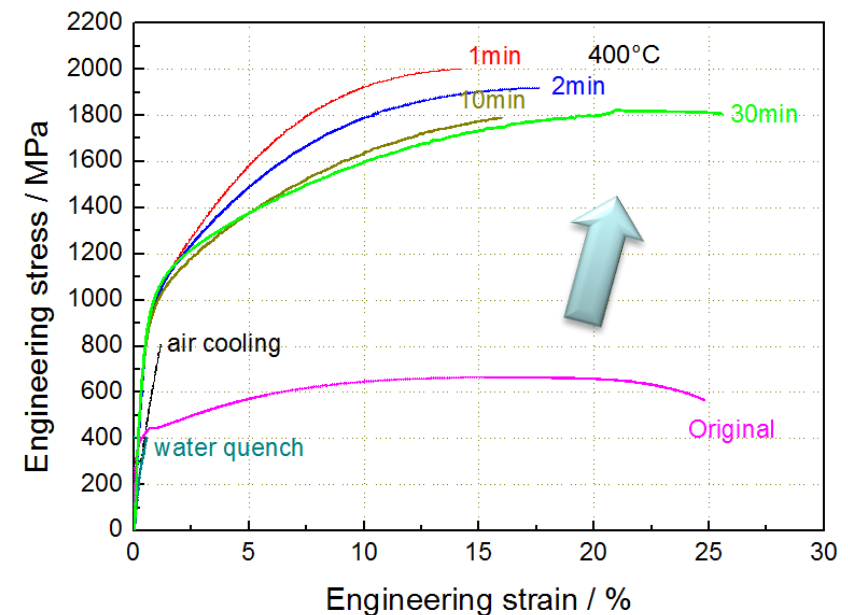
weldability

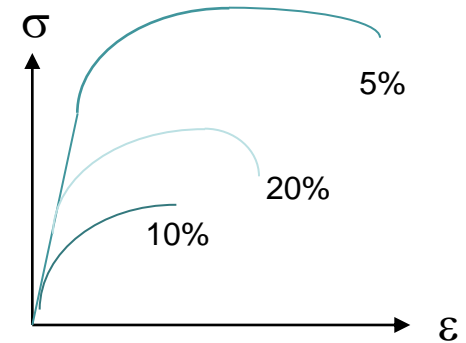
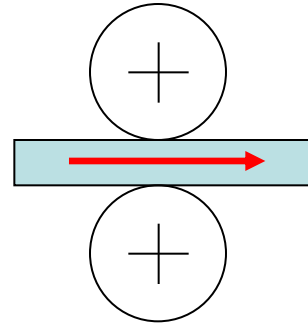
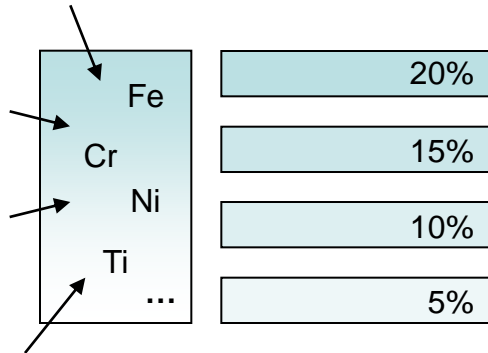
Conventional high-throughput methods do not work for structural materials:

correlation length scales relevant for mechanical properties > thin film thickness

correlation length scales created during processing, hence,

! SAME COMPOSITION ! but different processing





1. Generating of bulk specimens:

- Rapid change of (chemical) composition between single specimens
- Large variation of element-combinations (base materials)
- Fast / easy production (effort, speed, post processing)
- Few specific effects (e.g. porosity, dispersed oxides) desired

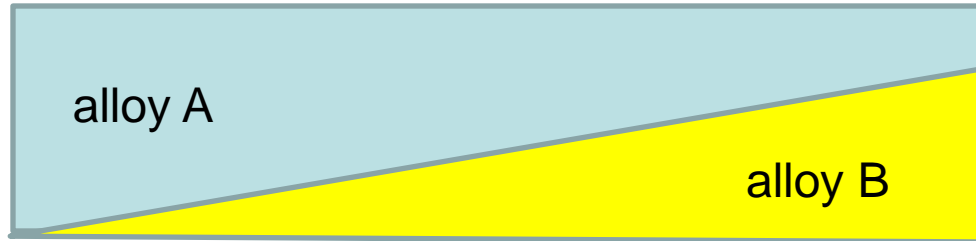
2. Thermomechanical treatment:

- Generate „appropriate“ microstructure representative for applications
- Ensure comparability of results to conventional methods
- Minimise / eliminate specific effects of generating method (cast microstructure etc.)

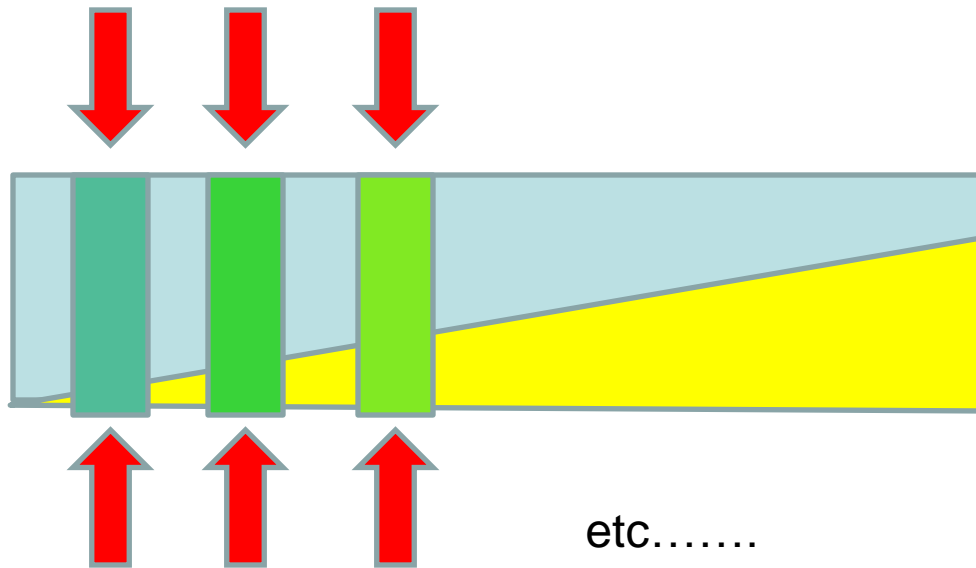
3. Material testing:

- Mainly tensile test for quick „fingerprinting“ (screening)
- Additional testing? (Flexible generating method for different sample dimensions)

step 1

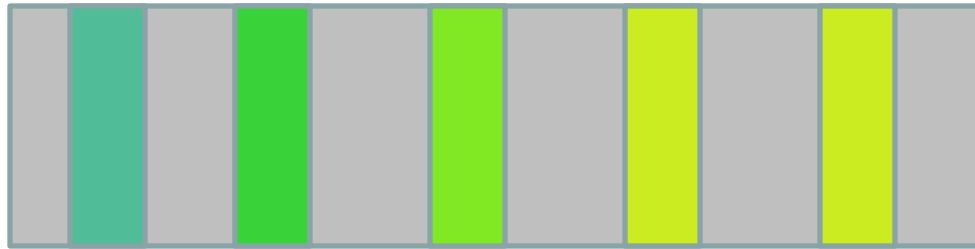


step 2



local melting
(inductive, LASER, electron beam, resistive welding, etc)

step 3

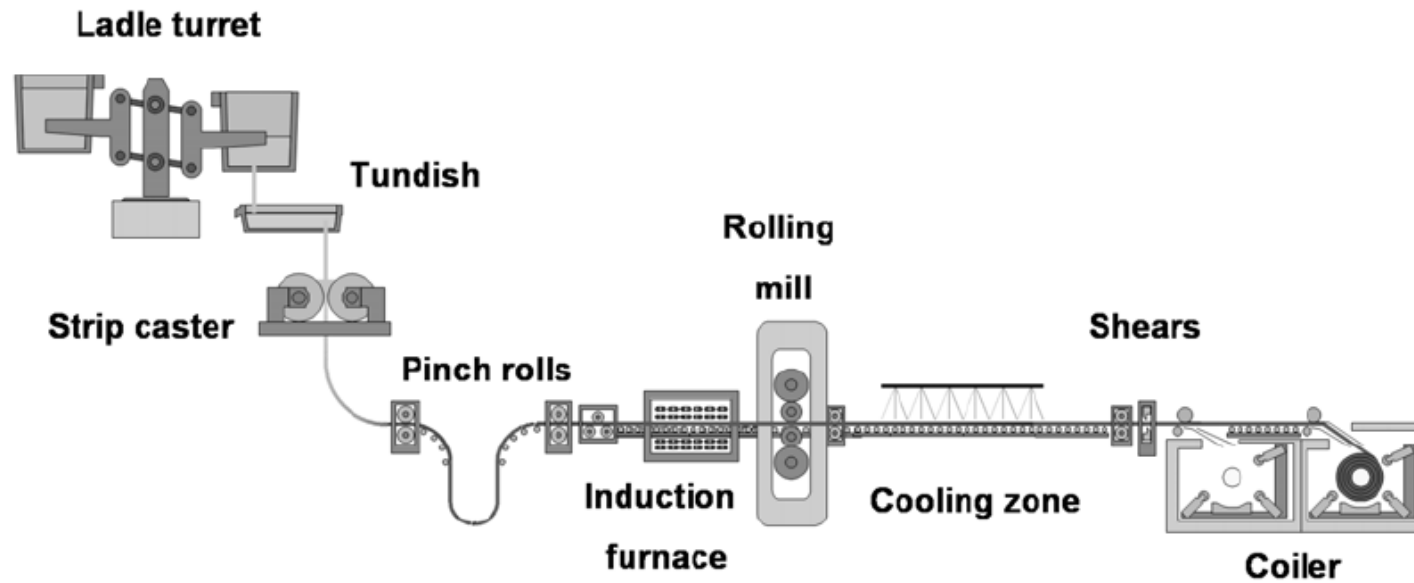


Thermomechanical treatment (e.g. rolling, annealing, etc: design microstructure after the solidified primary structure

step 4

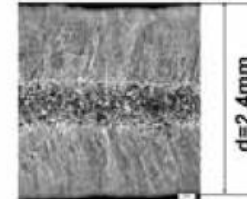


machining and mechanical testing (hardness not good enough: need tensile tests)



	<u>Slab casting</u>	<u>Thin strip casting</u>
Casting speed (m/min)	0,9 - 2,2	40 -100
Solidification rate (mm/s)	0,1	5
Heat flux density (MW/m ²)	1,0 - 2,5	5 -14
Solidification time (s)	10 ³	<0,5

Solidification structure
(AISI 304)



Rapid Alloy Prototyping – Development of High Throughput Methods

1. Laser-powdercoating (welding)

