

Nanoscale Transformations in Steels

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shortened version



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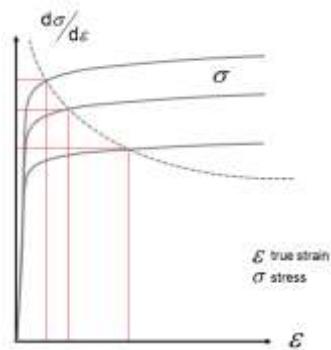
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13-16 April 2013
Shenyang, China

**Workshop on
Microstructure-driven
Design & Performance
of Advanced Metals**





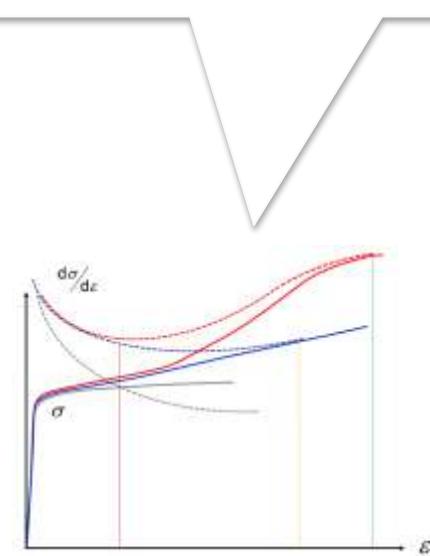
Inverse strength-ductility relation

Delayed onset of hardening:

Twins, martensite, dislocation substructures, gradual phase dissolution,...

Permanent strain hardening

Design strain hardening only where needed

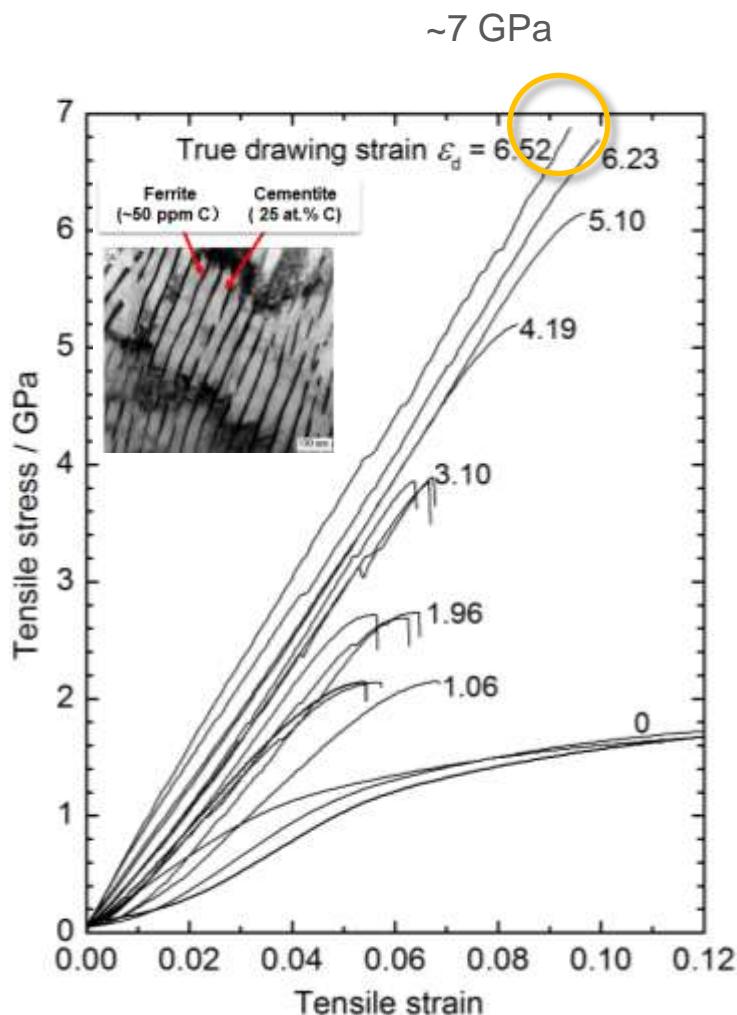




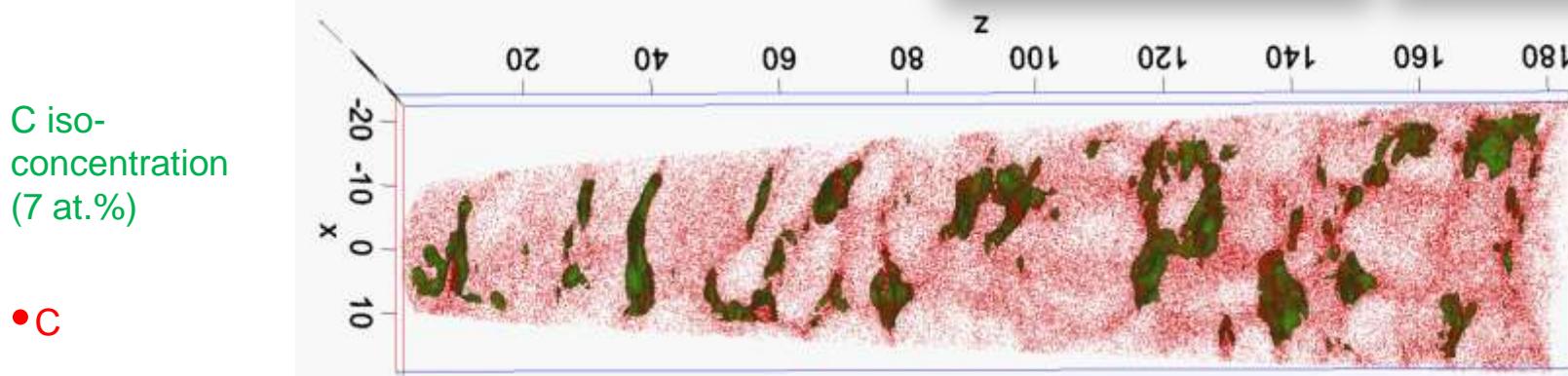
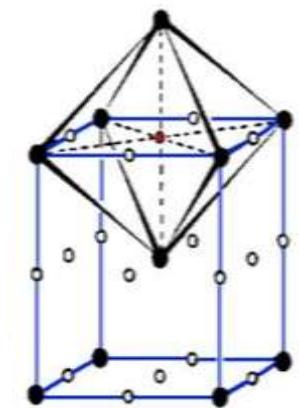
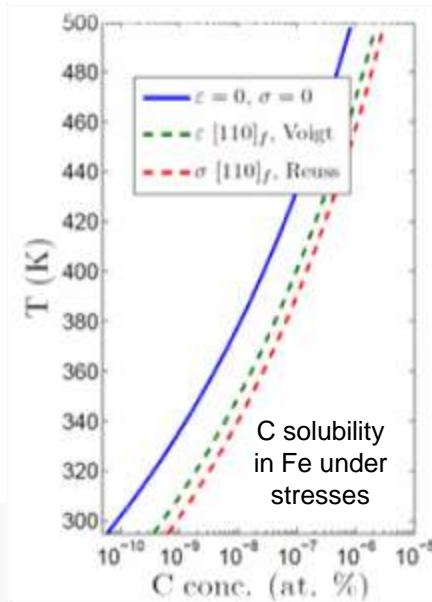
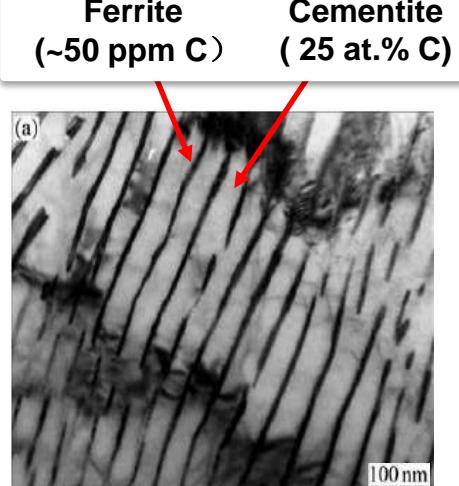
Pearlite: the limits of strength

Nano-austenite reversion

Fe-based superalloy



Towards the limits of strength: cold-drawn pearlitic steel



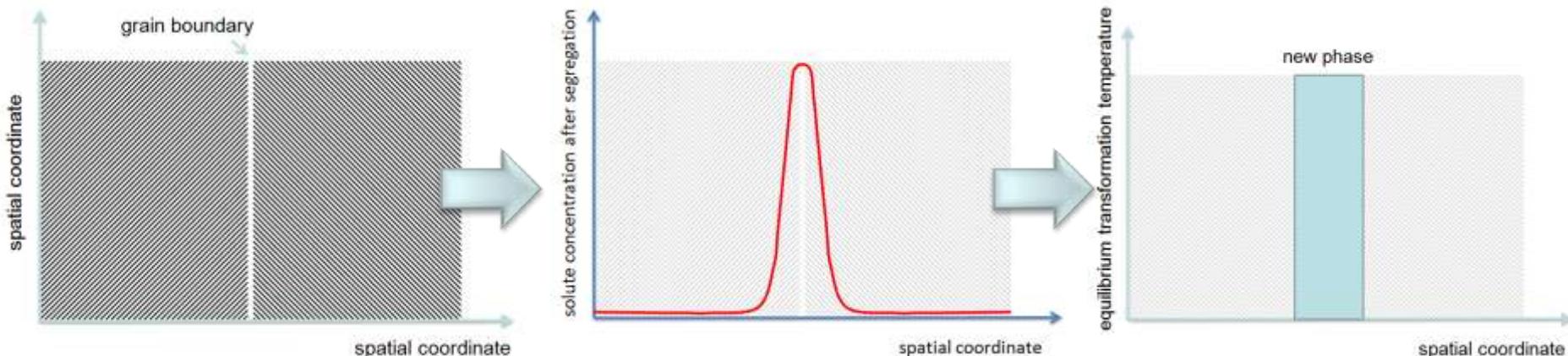
($\varepsilon = 6.5$)



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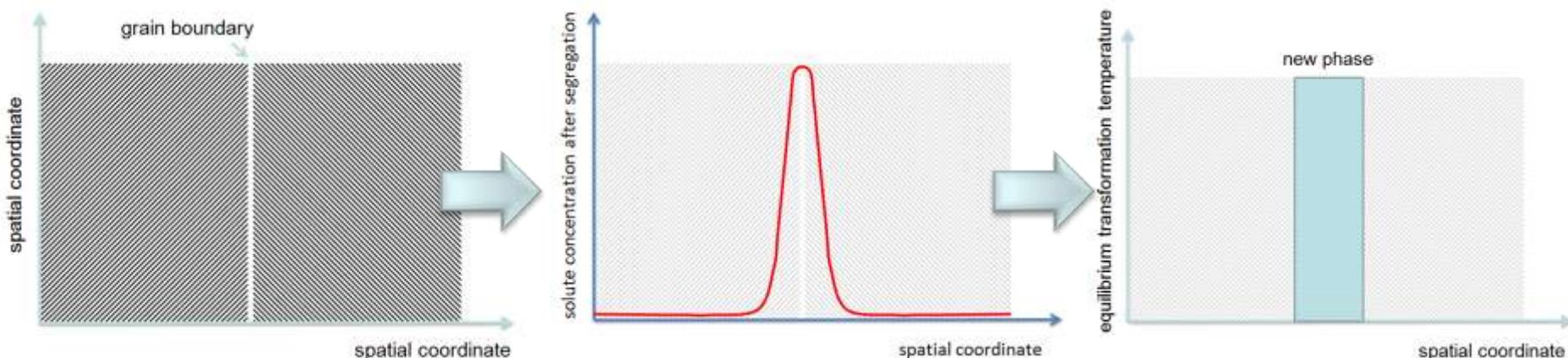
Fe-based superalloy



Solute segregation to martensite grain boundaries



Local phase transformation at grain boundary
(martensite-to-austenite reversion confined to GB)

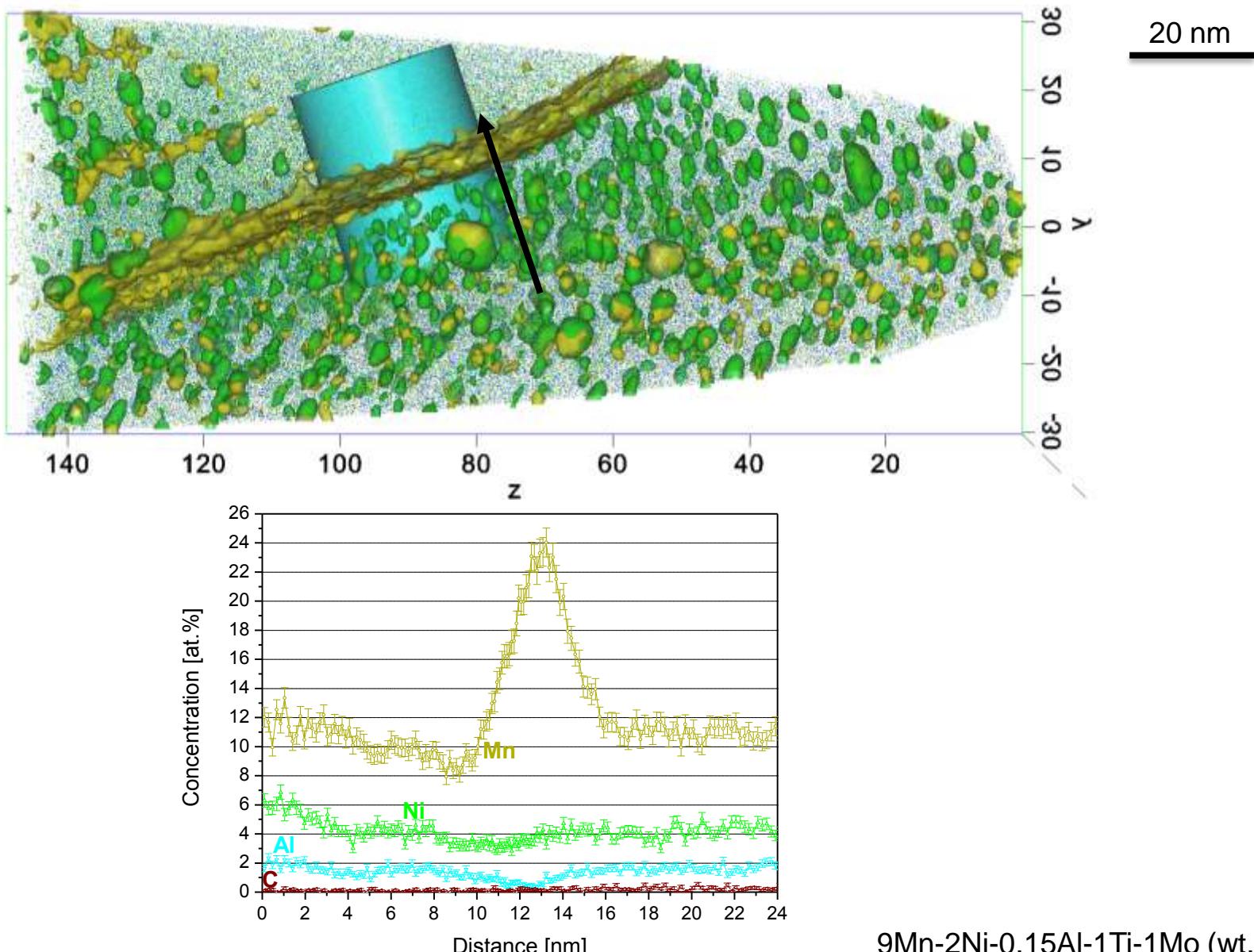


Solute segregation to martensite grain boundaries

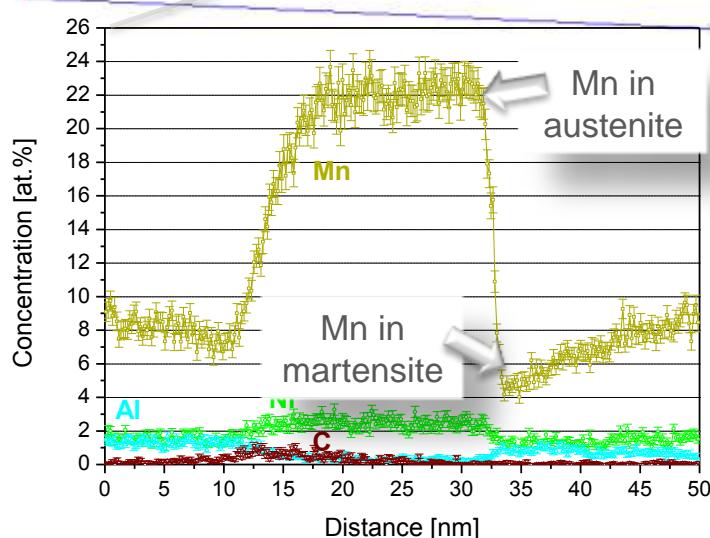
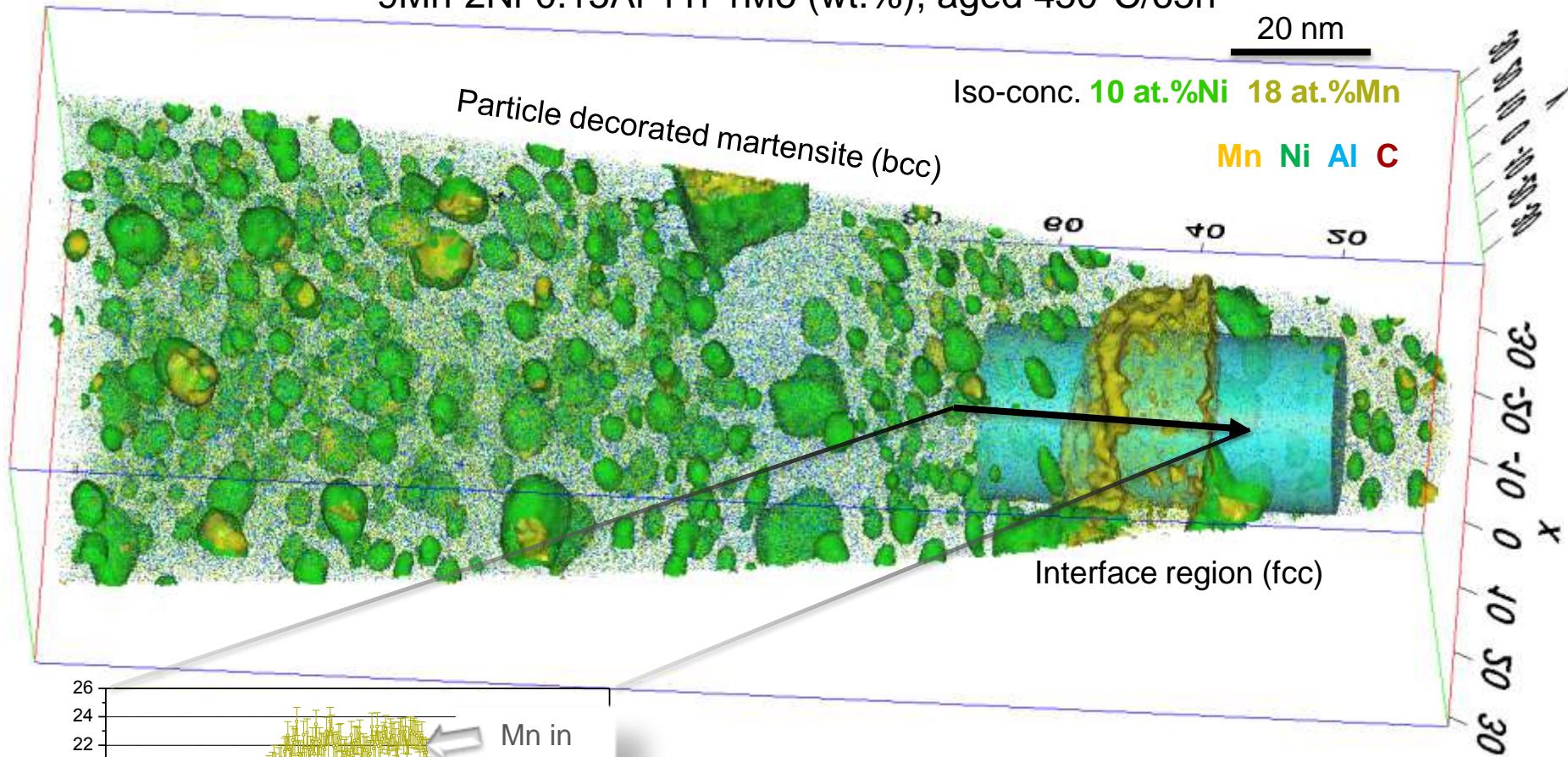
- Element with high segregation tendency
- Reduce transformation temperature (e.g. from martensite to austenite)
- Prefer segregation over bulk precipitation (e.g. carbide)

Local phase transformation at grain boundary
(martensite-to-austenite reversion confined to GB)

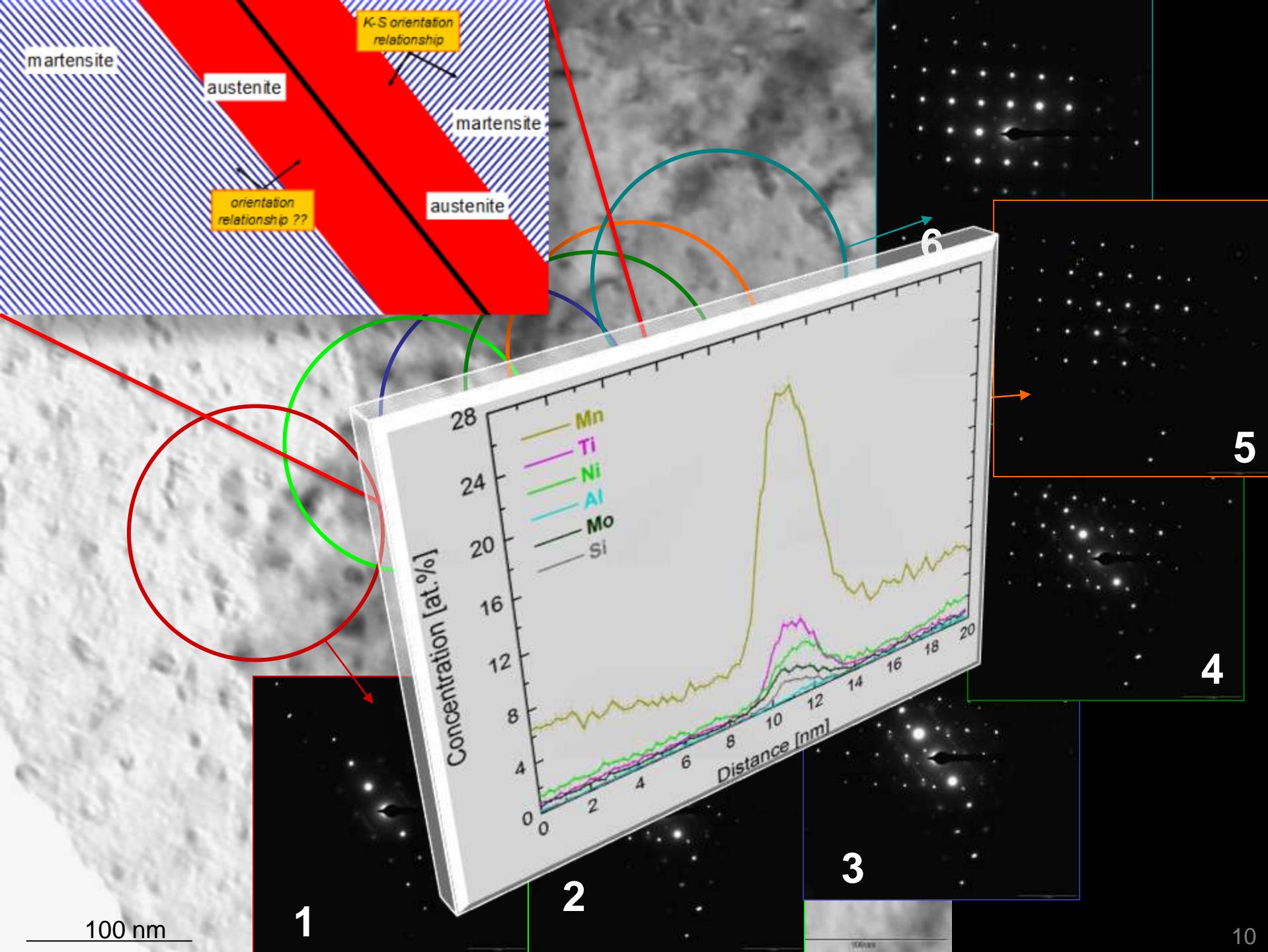
Mn segregation at grain boundary, (450°C/65h)



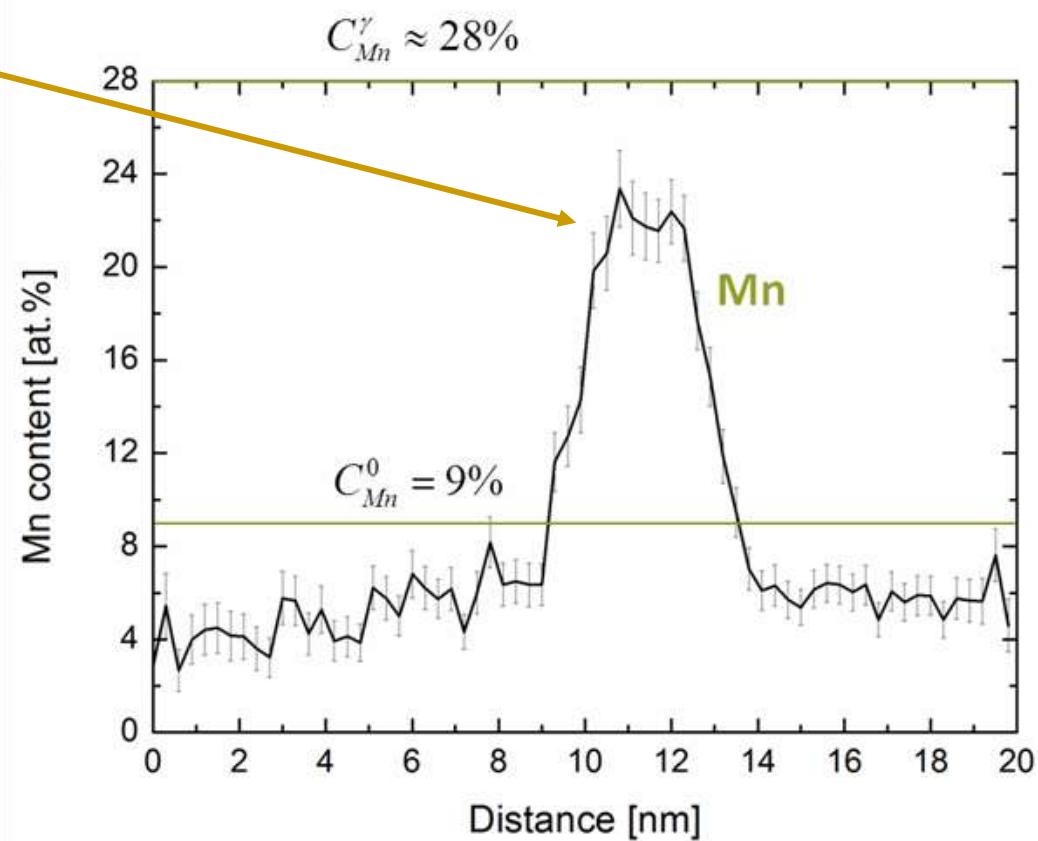
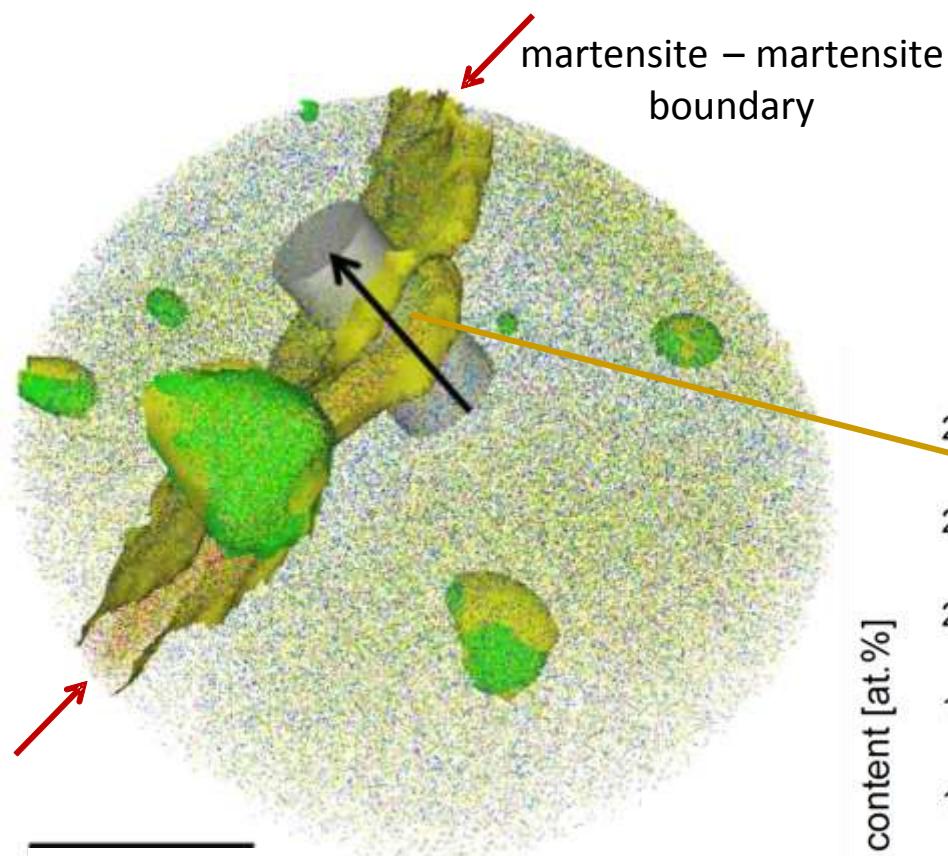
20 nm



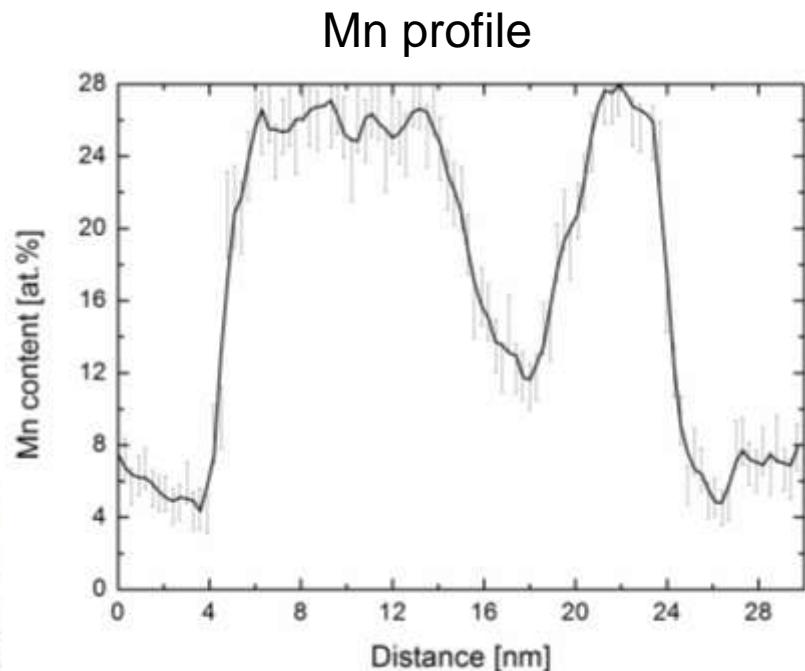
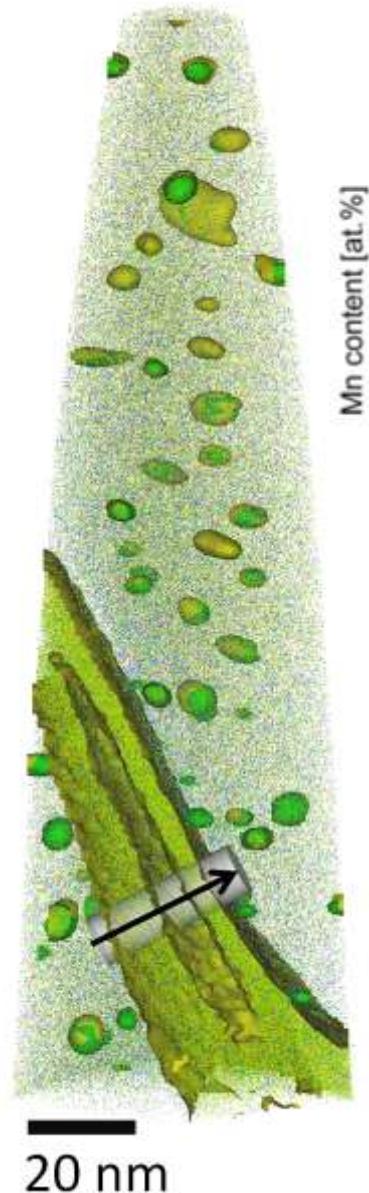
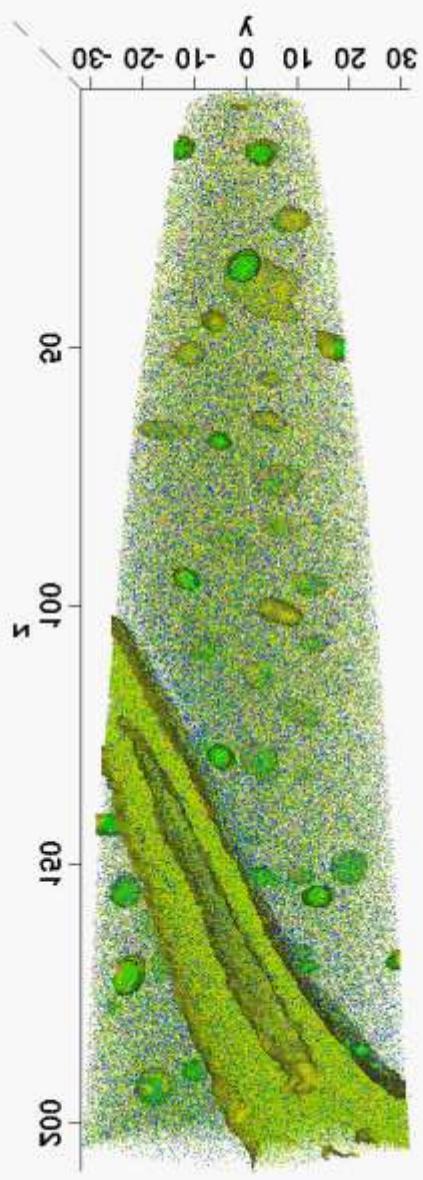
Phase formation at martensite interface
Near-equilibrium partitioning at interface



Thin intergranular FCC layer among two martensite crystals, APT



Growth of retained austenite 450°C/48h. 12MnPH

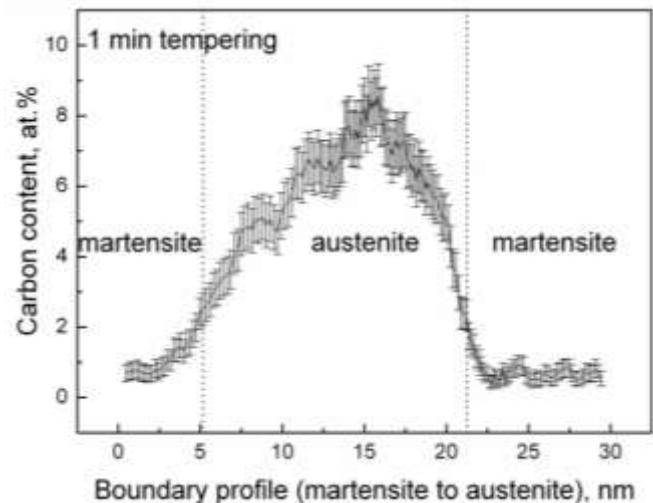
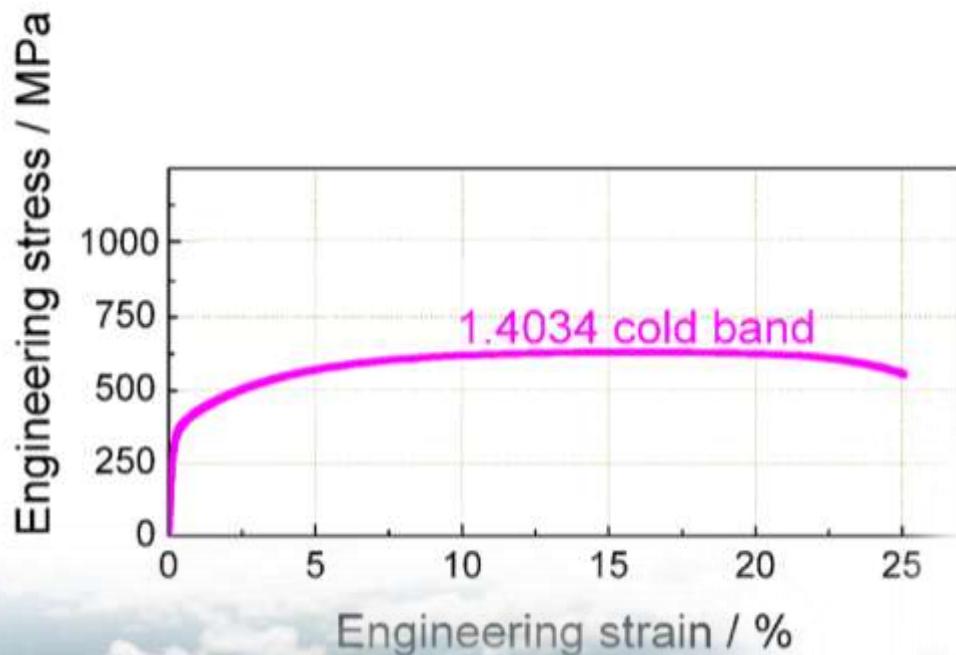


- Mn atoms
- Ni atoms
- Al atoms
- Ti atoms
- Mo atoms
- Fe atoms
- 10at.% Ni iso-conc. surface
- 22at.% Mn iso-conc. surface

C	Ni	Mo	Ti	Al	Mn	Fe
0.01	2.0	1.0	1.0	0.15	12	bal.

650 MPa to 2 GPa

Making martensite ductile





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Nano-austenite reversion

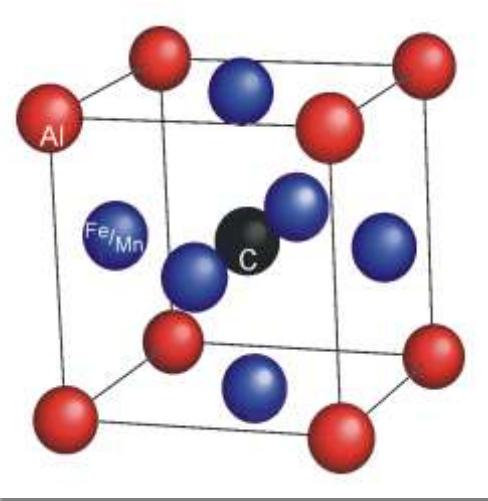
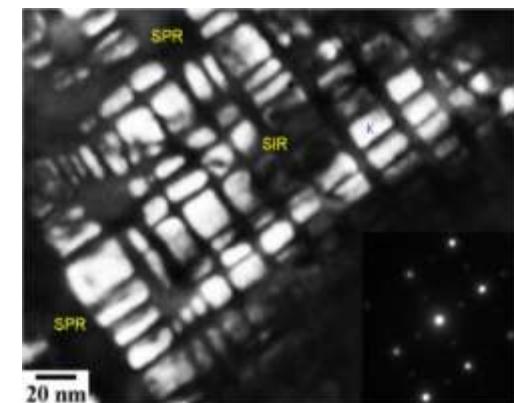
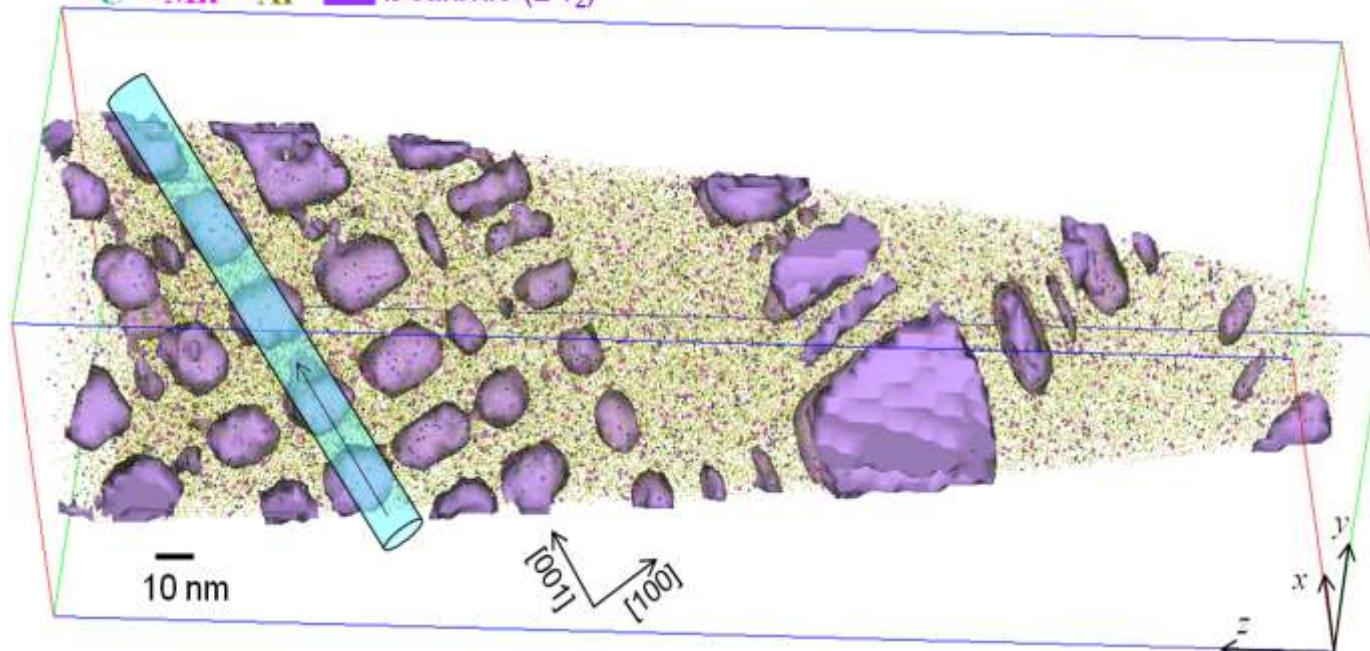
Fe-based superalloy

Fe-Mn steel, weight reduced (10% less mass density)



•C •Mn •Al ■ κ -carbide ($L'1_2$)

κ -carbide (iso-surface of 9 at.% C)



Lattice structure
of κ -Carbides,
Perowskite type

- Design alloys by self-organized nanostructuring
- Segregation plus confined phase transformation at defects
- Works for dislocations too?
- Deformation-driven mechanical bulk alloying leads to non-equilibrium phases approaching the theoretical limits of strength
- Designing stable nanocarbides enables weight-reduced ultra-ductile and thermally stable materials

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