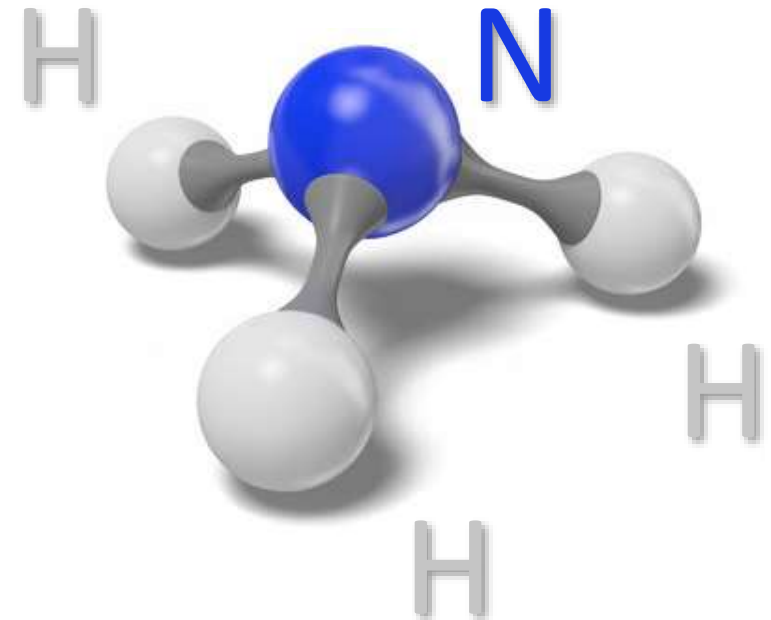


Green steel with Hydrogen from Ammonia (NH_3)



What is



'Green Steel'?

D. Raabe, The Materials Science behind Sustainable Metals and Alloys, Chem. Rev. 2023, 123, p. 2436

OPEN ACCESS - Free LINKS:

<https://pubs.acs.org/doi/full/10.1021/acs.chemrev.2c00799>

<https://onlinelibrary.wiley.com/doi/full/10.1002/advs.202300111>

<https://www.nature.com/articles/s41586-019-1702-5>

Steel with...

lower carbon footprint

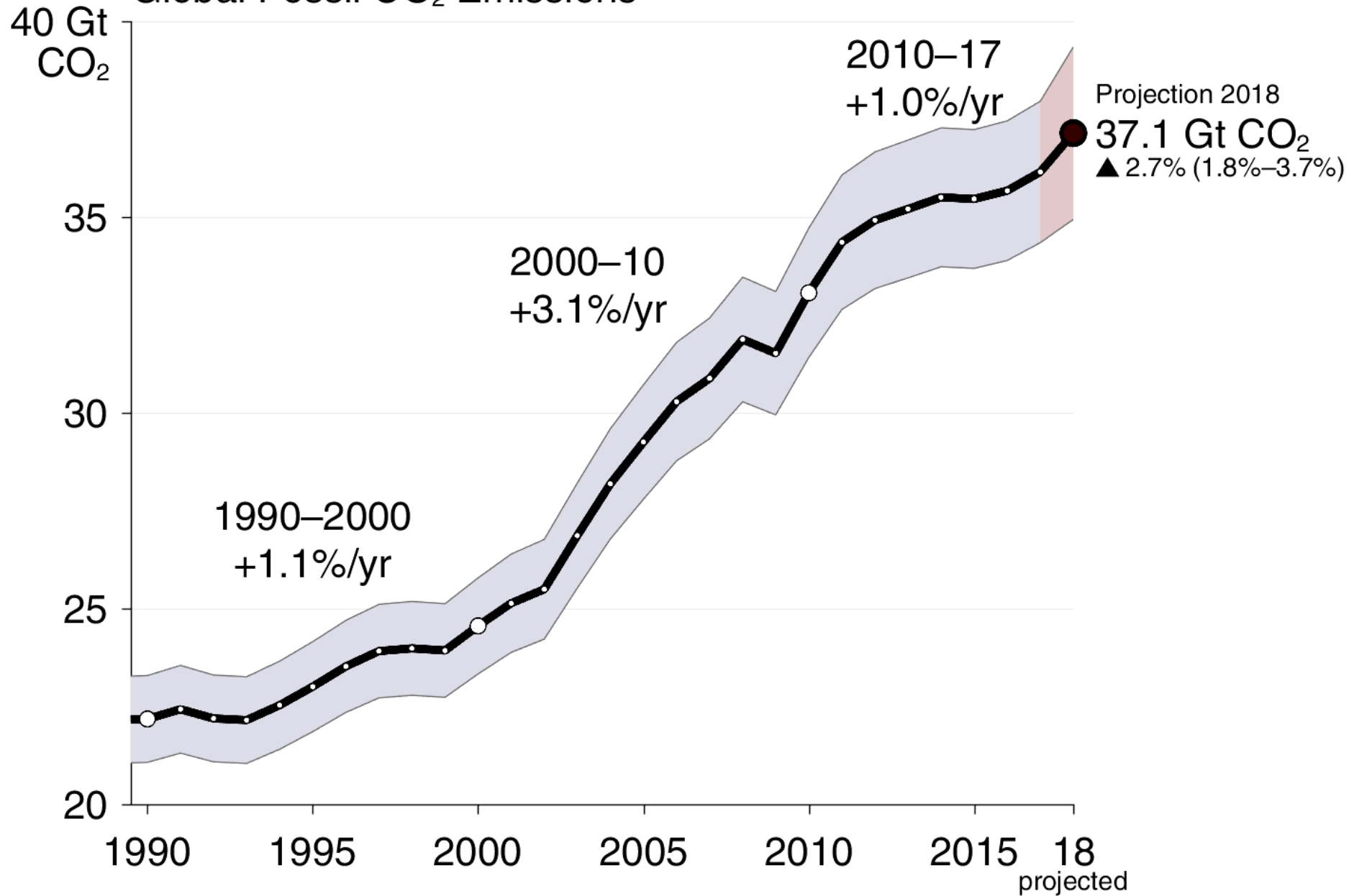
lower energy use

less byproducts

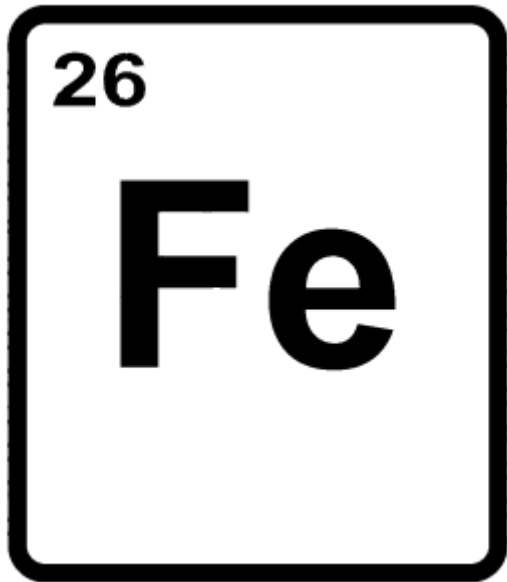
higher efficiency



Global Fossil CO₂ Emissions



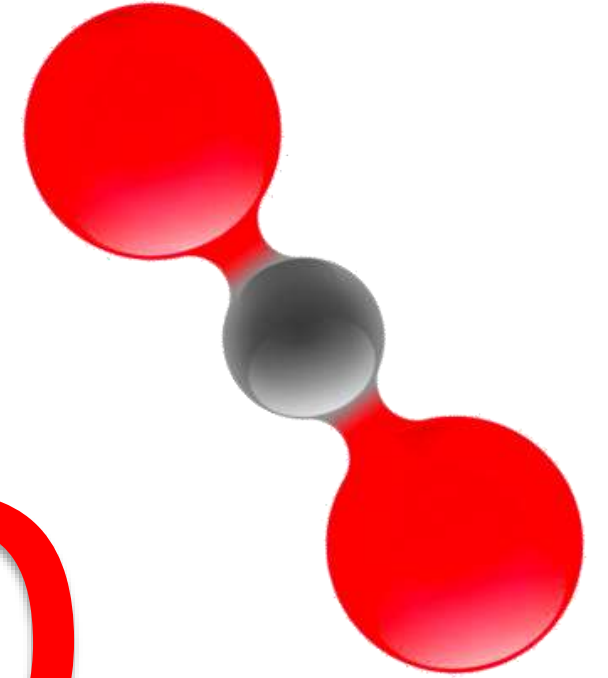
Steel: 8 % of all global
 CO_2 emissions



Today:

2 tons CO₂

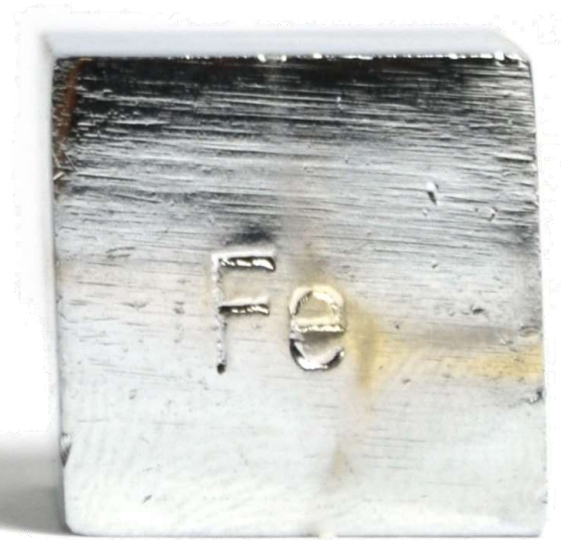
per ton of steel



Why has steel
such a high
carbon footprint

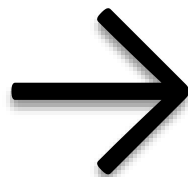


Redox reaction

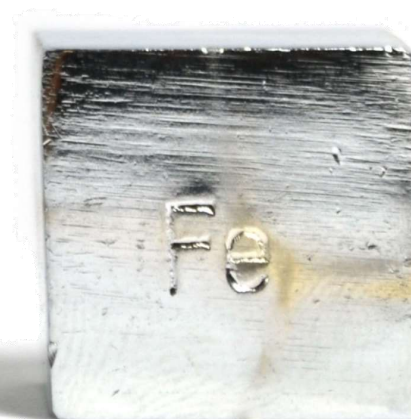
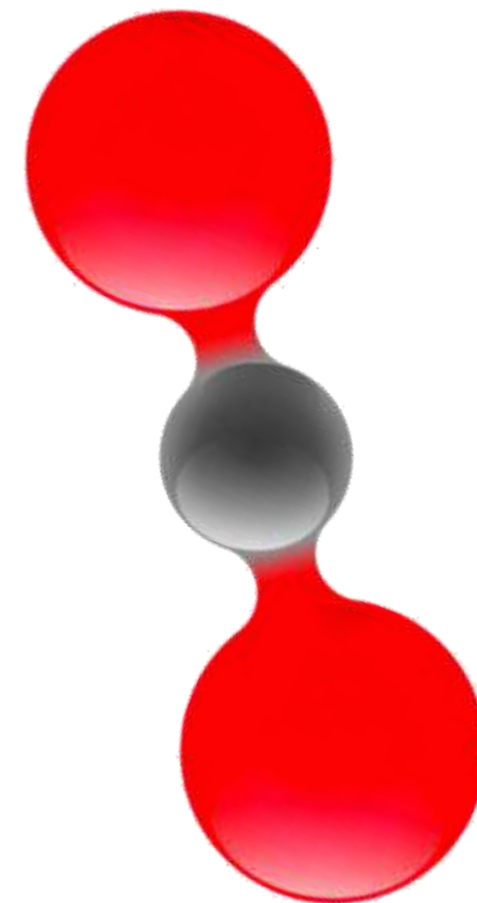




+



+



The Materials Science behind Sustainable Metals and Alloys

Dierk Raabe*



Cite This: *Chem. Rev.* 2023, 123, 2436–2608

Open
Access



Review

Strategies for improving the sustainability of structural metals

<https://doi.org/10.1038/s41586-019-1702-5>

Dierk Raabe^{1*}, C. Cem Tasan^{2*} & Elsa A. Olivetti^{2*}

1.9 billion tons steel / year

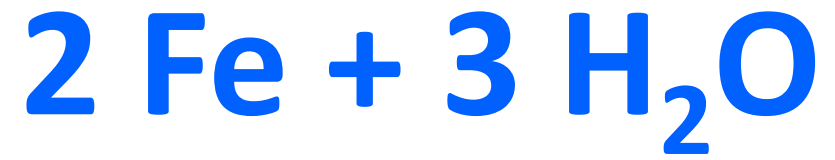
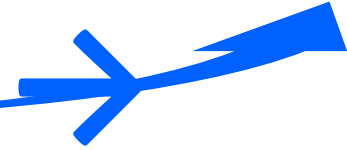
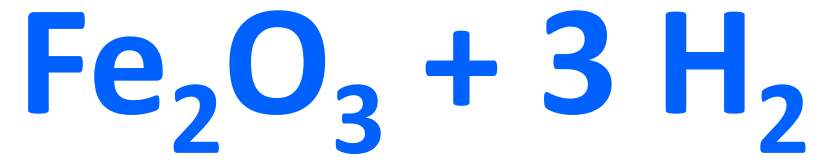




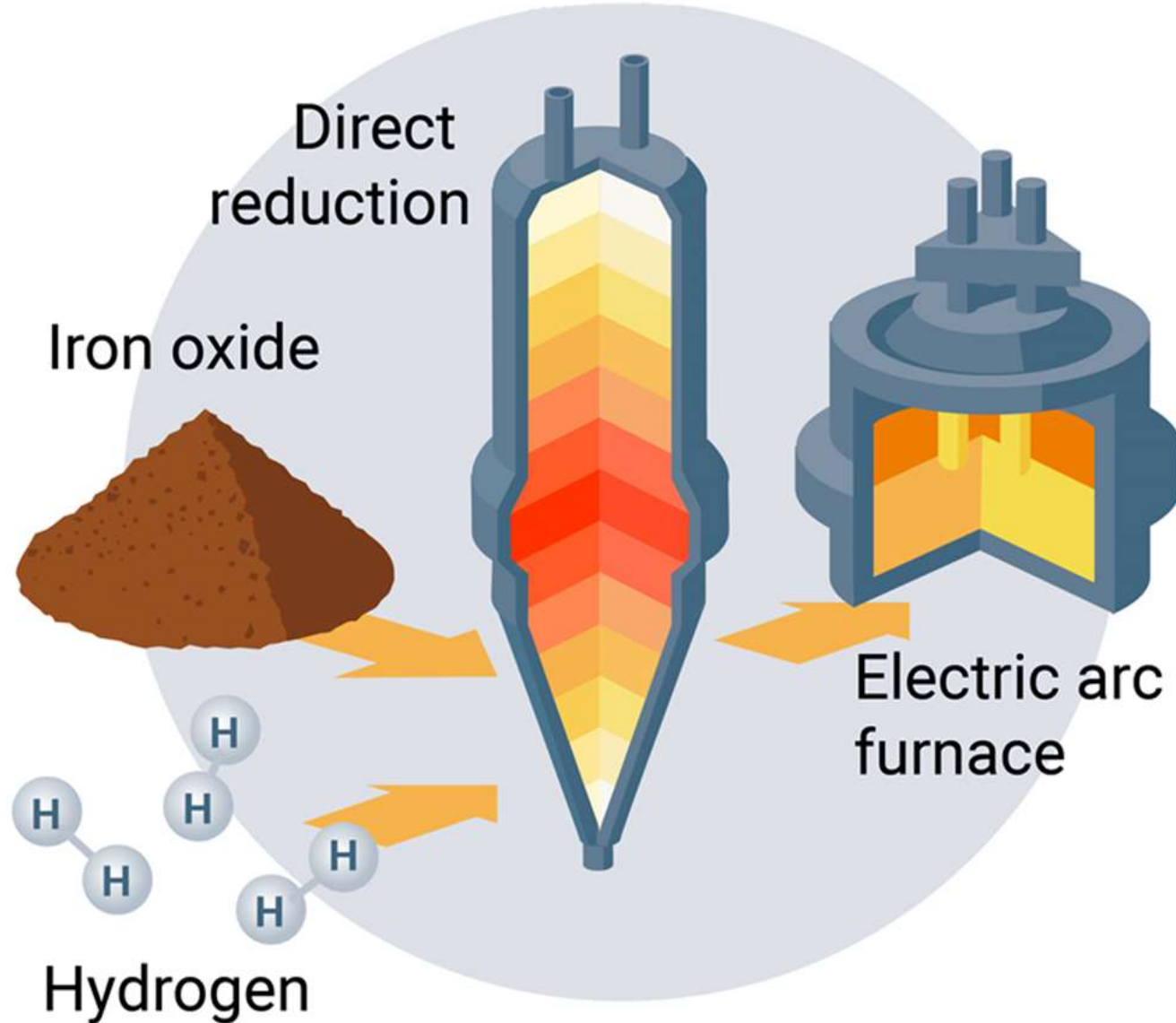
2.5 billion tons



1.9 billion tons



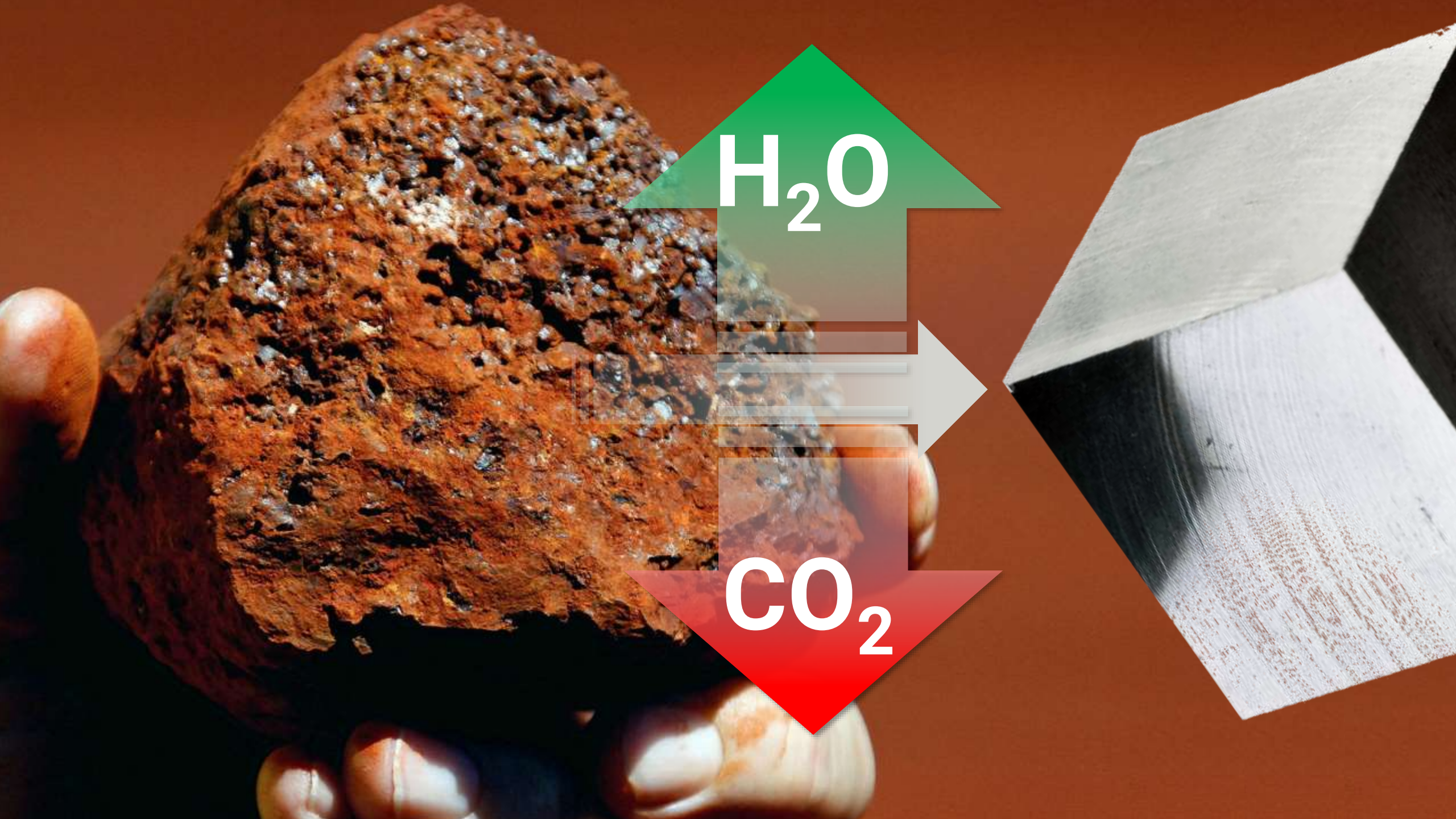
Direct reduction & electric arc 70-80% less CO₂ when using H₂





Handwritten text at the top of the page, possibly a title or date, which is mostly illegible due to blurring.





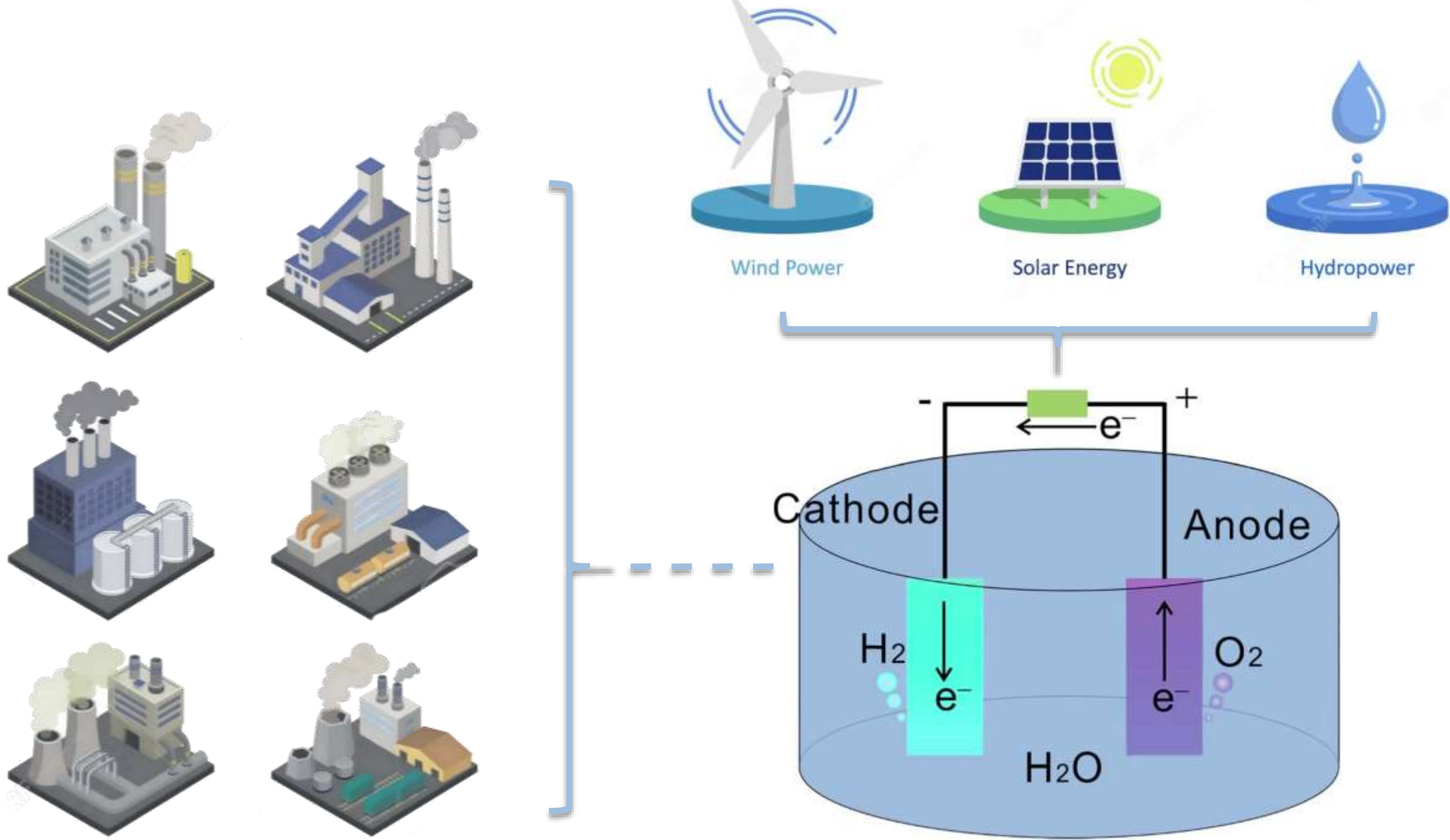
H_2O

CO_2

Where does all the
green hydrogen
come from?

(>250 Million tons H₂ required)





Producing 1 kg hydrogen,
which has a specific energy of
143 MJ/kg, requires 50-55 kW
h (180- 200 MJ/kg)

Water splitting by polymer
electrolyte membrane or
alkaline electrolysis:

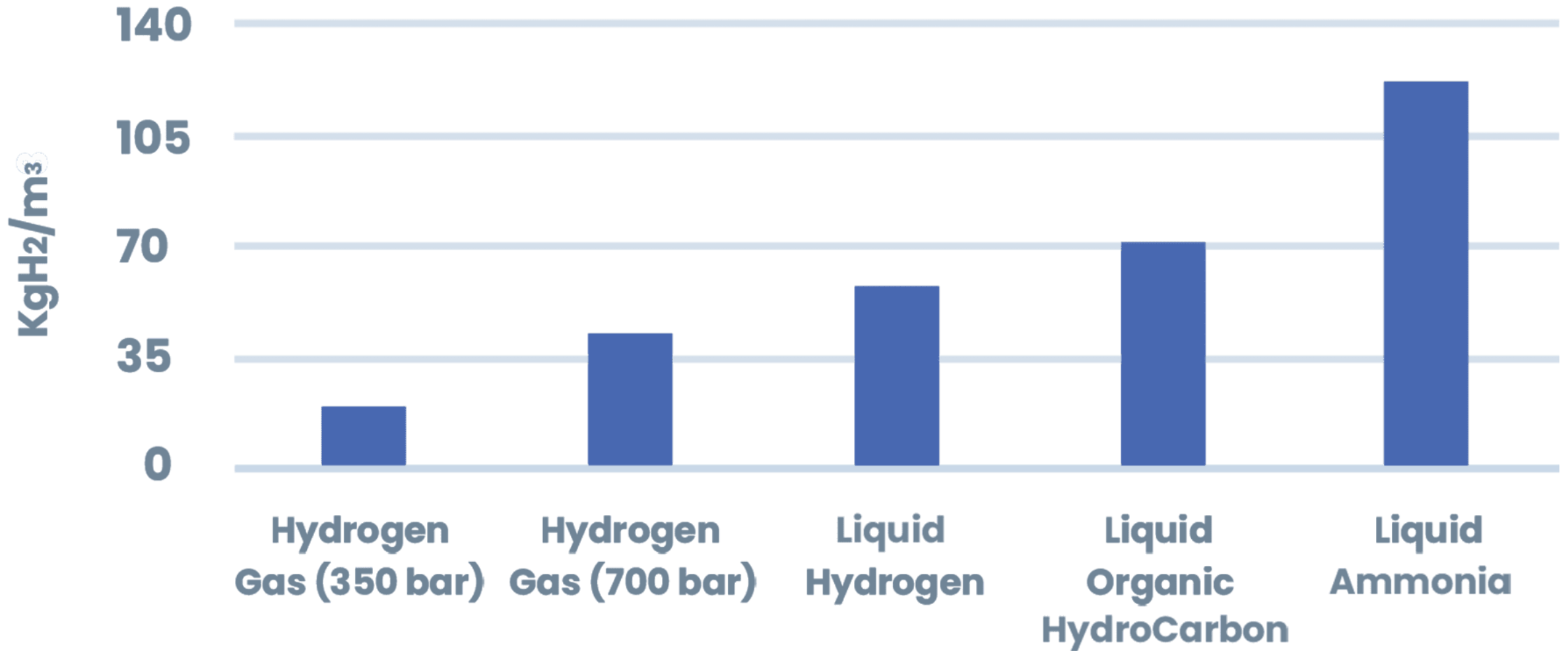
70-80% efficiency



Liquefaction to
-253°C costs 35%
of its energy



Hydrogen density per m³



Liquefaction of hydrogen:
consumes 35% of its energy

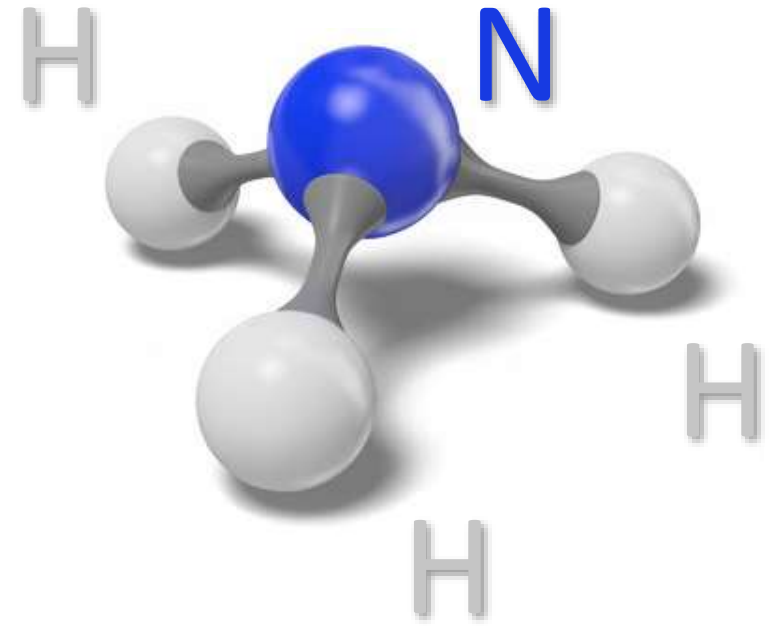
i.e.

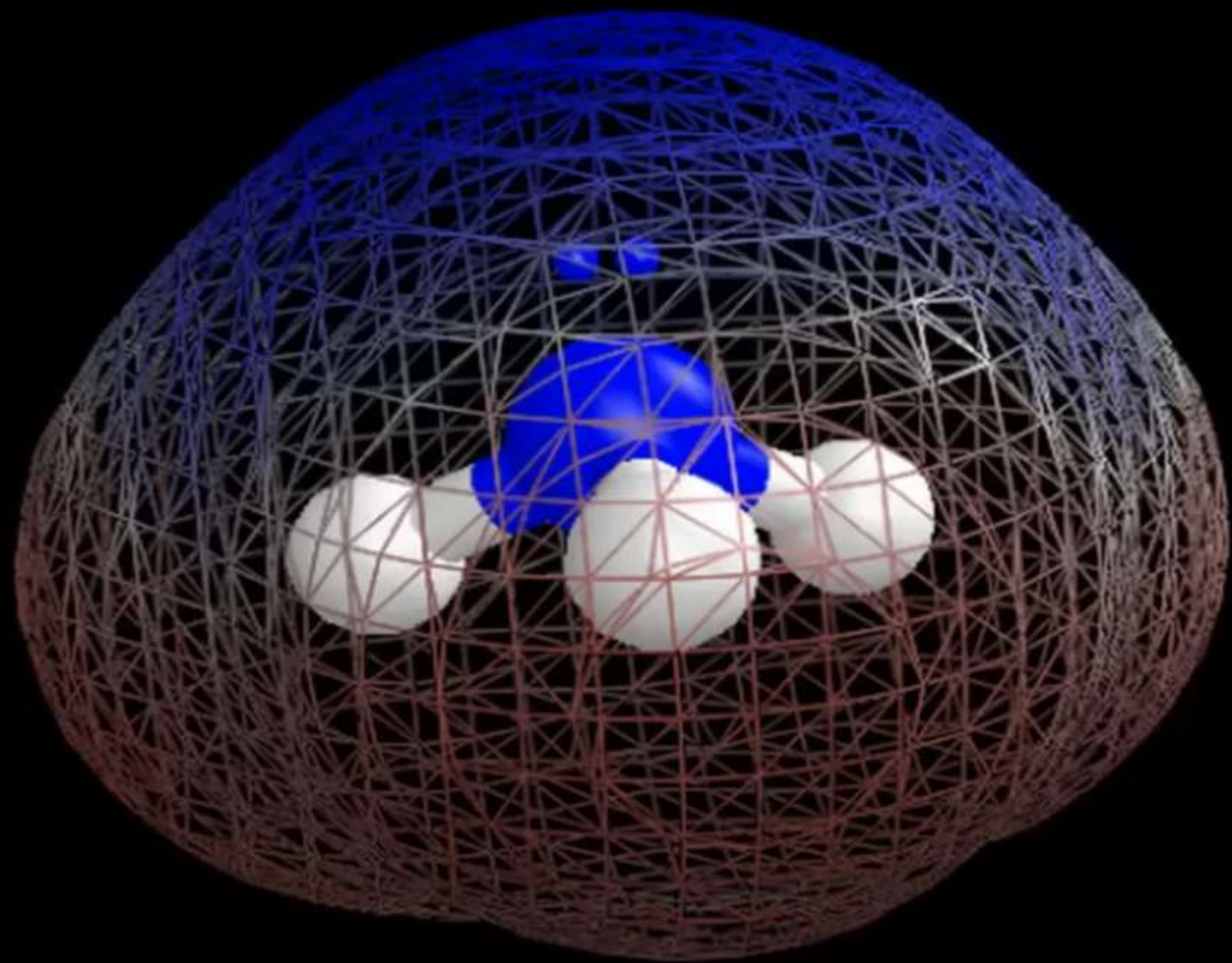
liquid hydrogen transport
is inefficient !

Is Ammonia

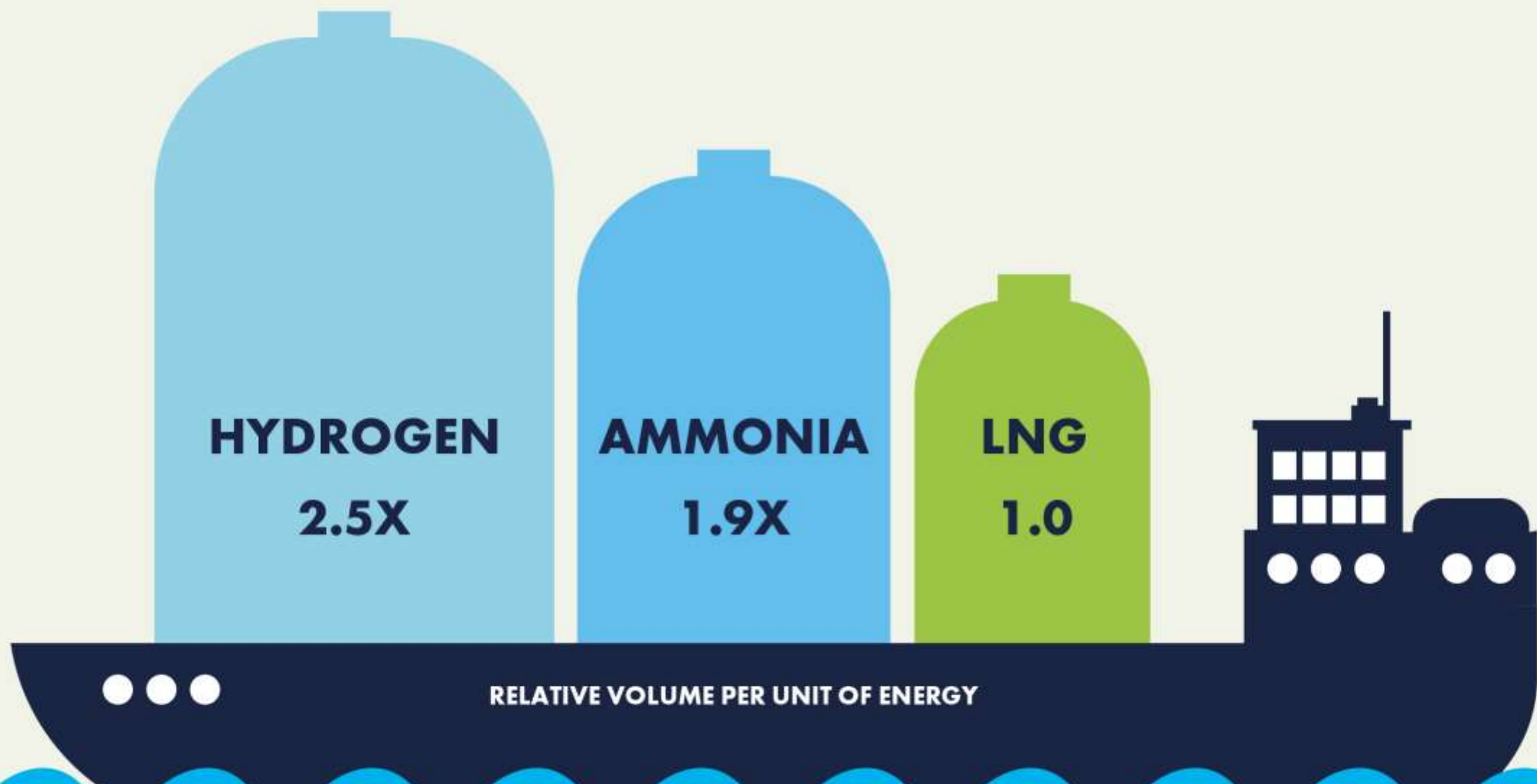
an alternative / vector to

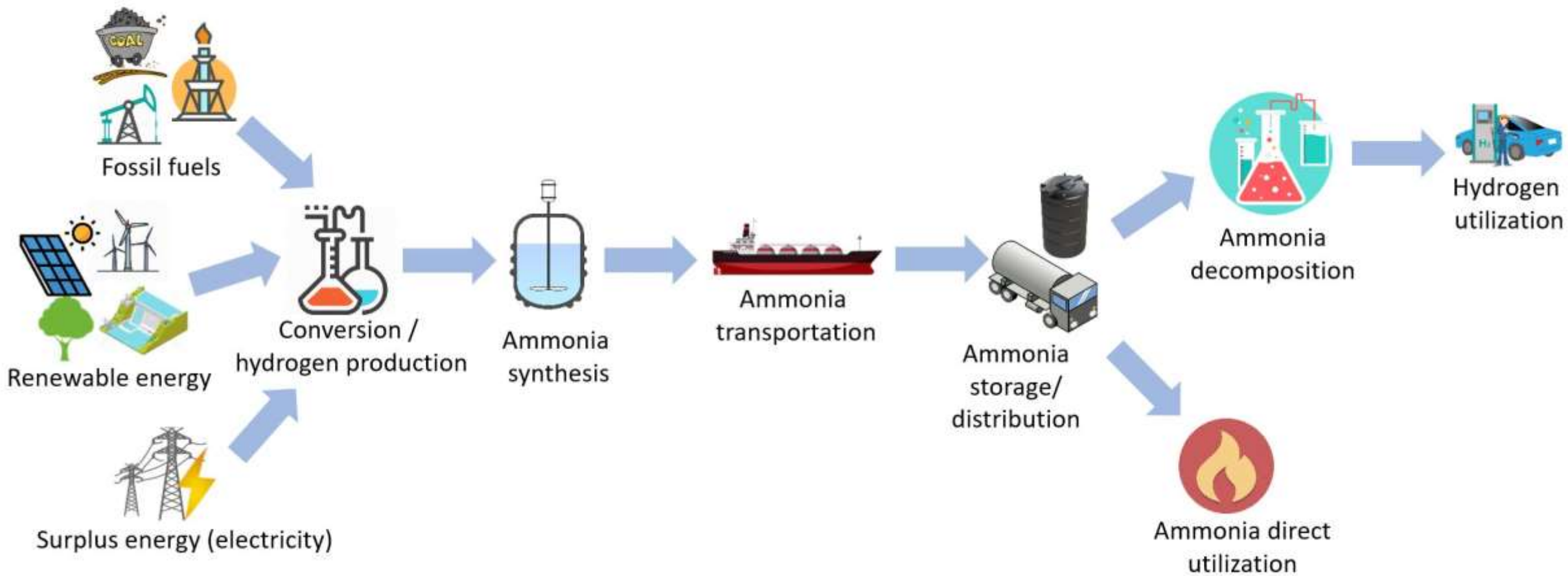
Hydrogen?



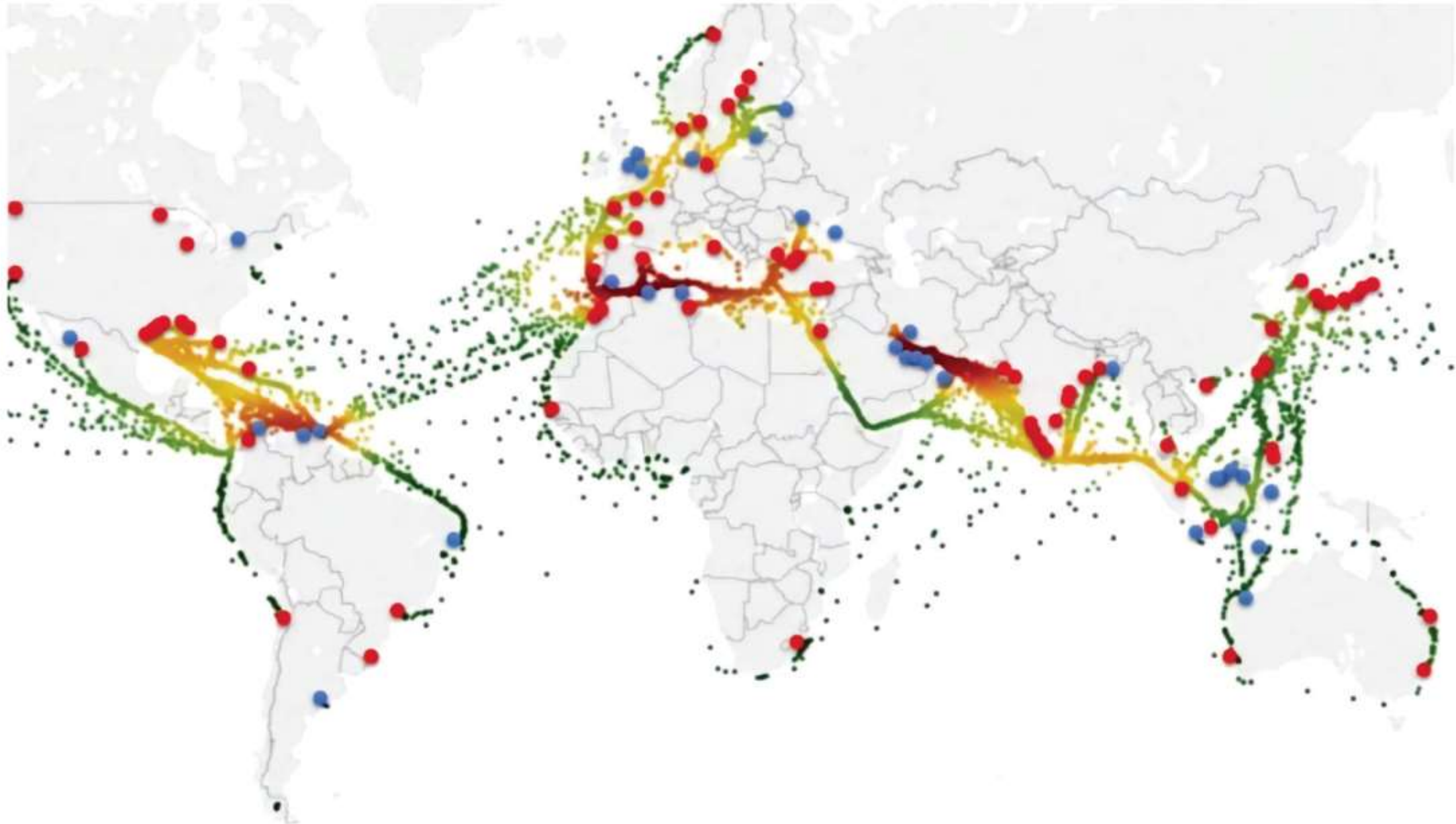


TANKS SIZE MATTERS



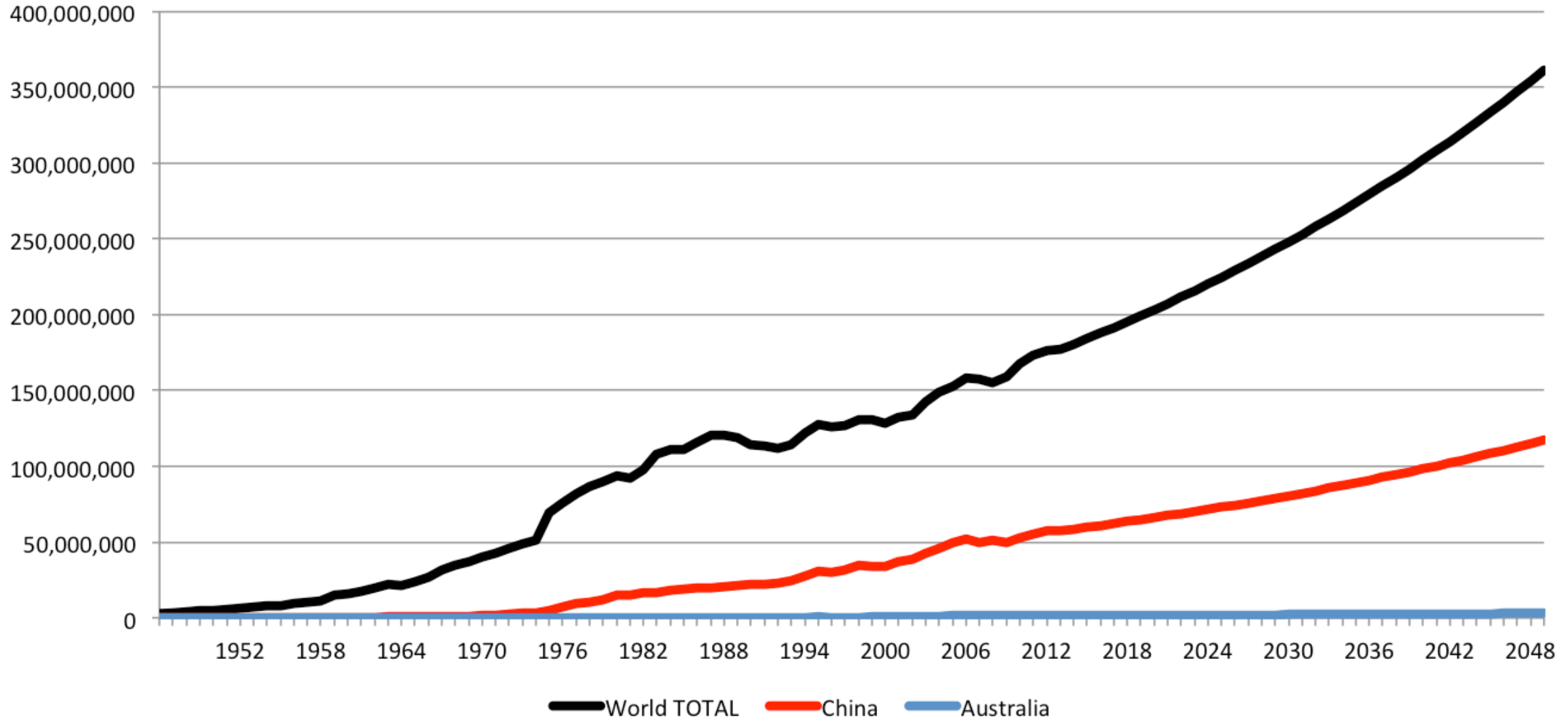


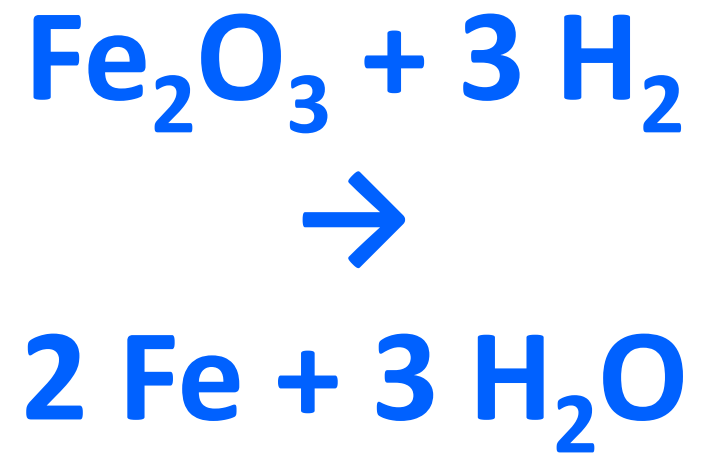
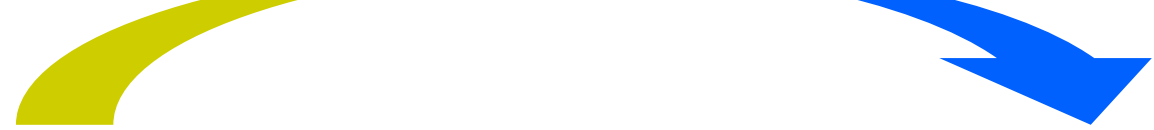
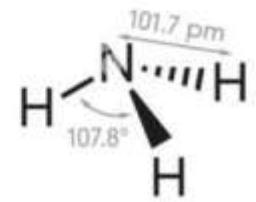
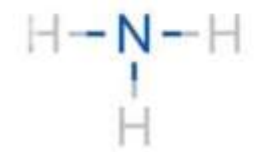
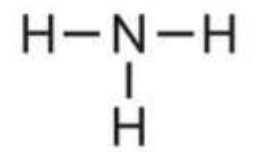
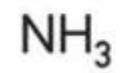
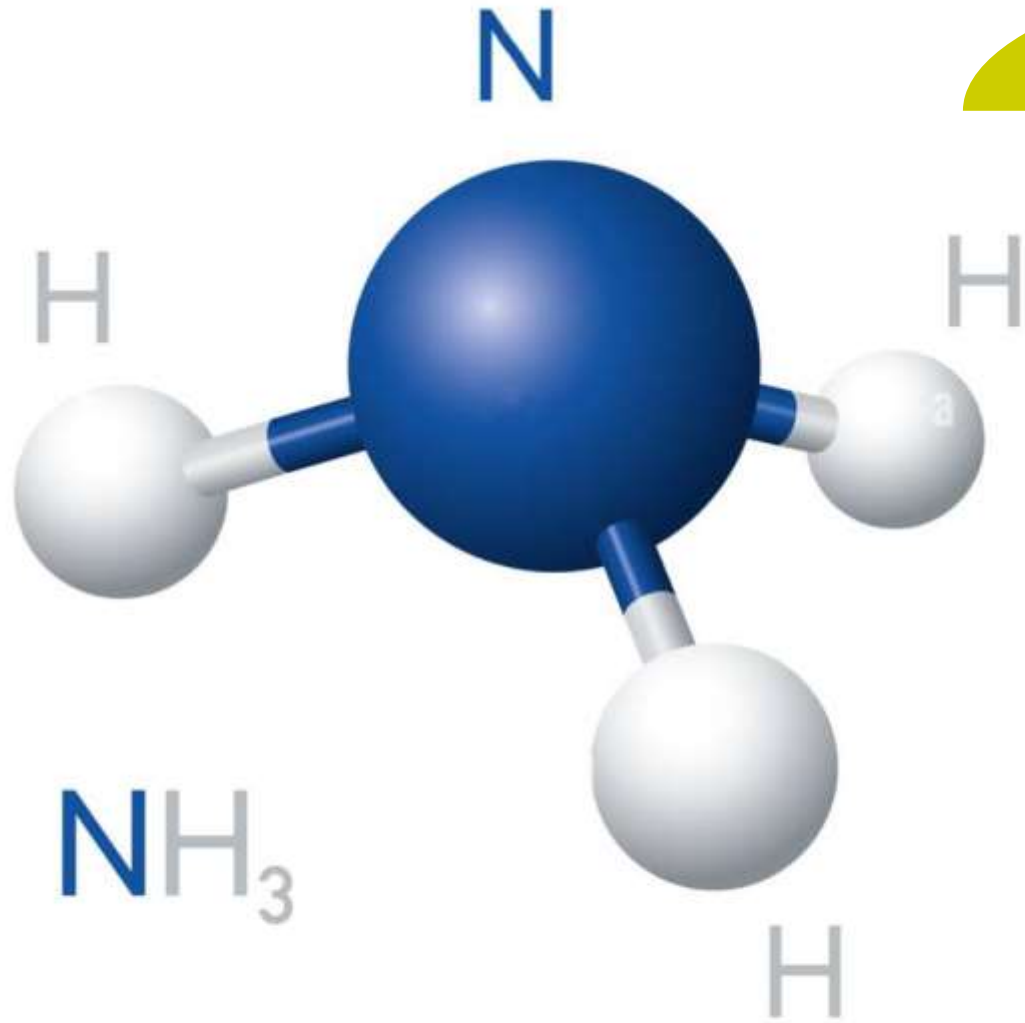
● Ammonia loading facilities ● Ammonia unloading port facilities



Global Ammonia Production, forecast to 2050 @ 2% growth

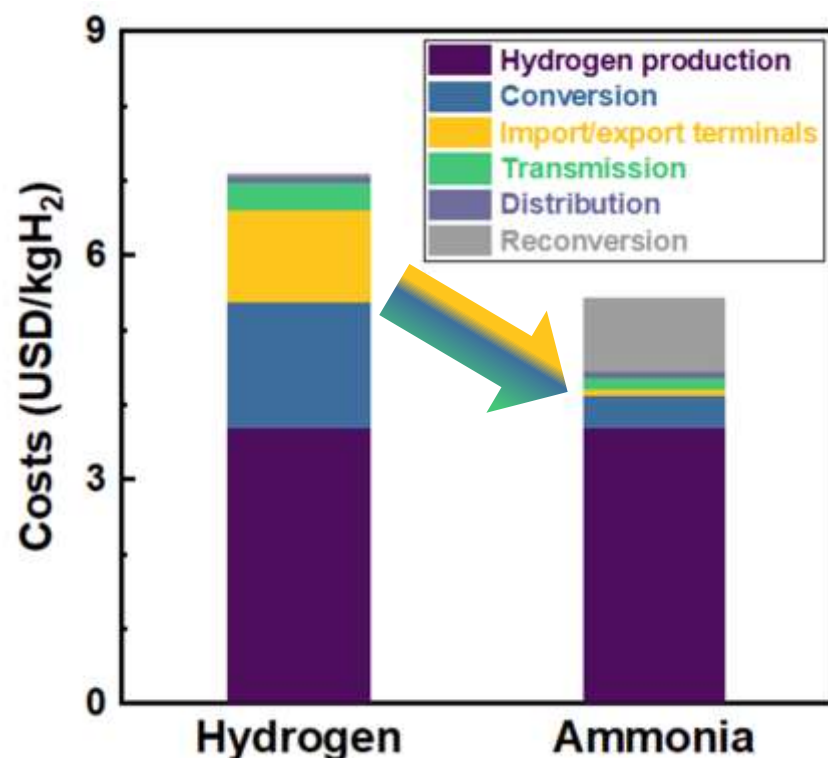
Units: metric tons of ammonia. Source: US Geological Survey (1947-2014) / AmmoniaIndustry.com, March 2018





Reducing Iron Oxide with Ammonia: A Sustainable Path to Green Steel

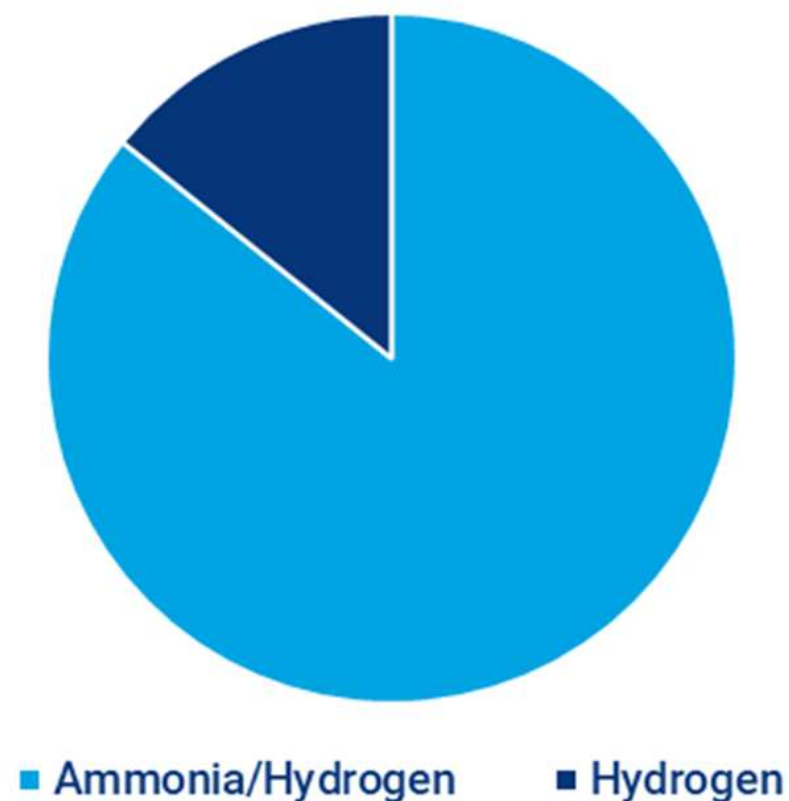
Yan Ma,* Jae Wung Bae, Se-Ho Kim, Matic Jovičević-Klug, Kejiang Li, Dirk Vogel, Dirk Ponge, Michael Rohwerder, Baptiste Gault, and Dierk Raabe*



... renewable
carbon-free reductants and electricity. Here, the authors show how to make sustainable steel by reducing solid iron oxides with hydrogen released from ammonia. Ammonia is an annually 180 million ton traded chemical energy carrier, with established transcontinental logistics and low liquefaction costs. It can be synthesized with green hydrogen and release hydrogen again through the reduction reaction. This advantage connects it with green iron making, for replacing fossil reductants. The authors show that ammonia-based reduction of iron oxide proceeds through an autocatalytic reaction, is kinetically as effective as hydrogen-based direct reduction, yields the same metallization, and can be industrially realized with existing technologies. The produced iron/iron oxide mixture can be frequently melted in electric

Energy density is a key factor in ammonia's role as a hydrogen carrier

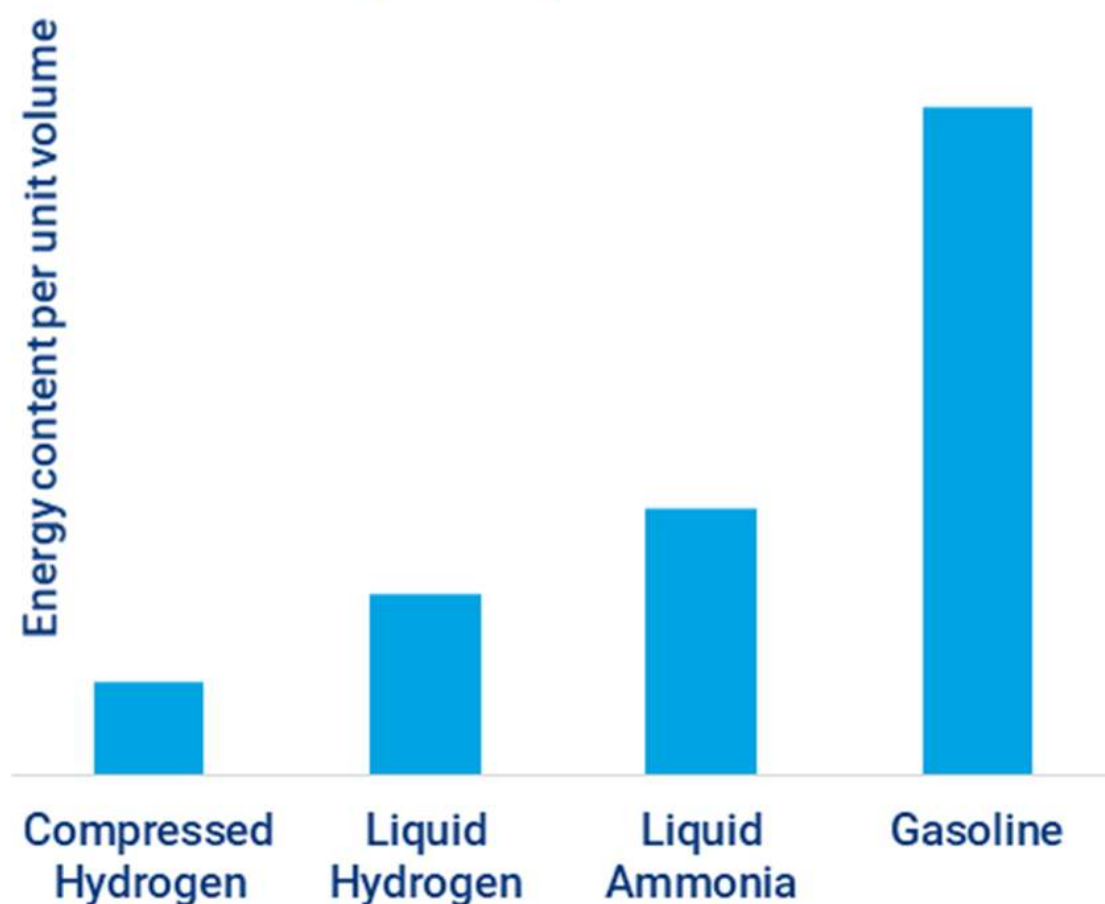
Announced hydrogen projects by capacity*



*In Australia, the Middle East, Africa and Latin America.

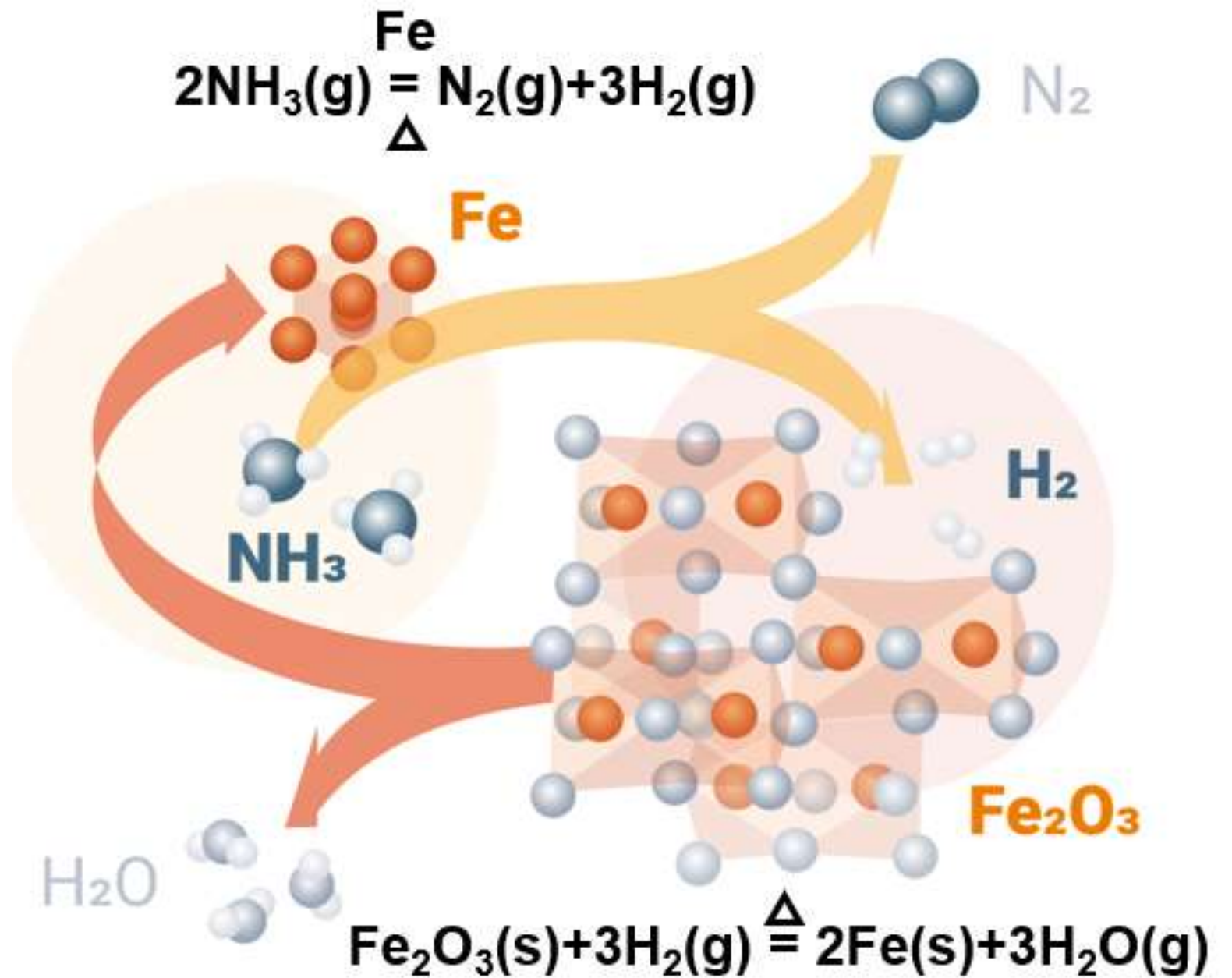
Source: Wood Mackenzie Hydrogen Project Tracker

Relative energy density of selected fuels

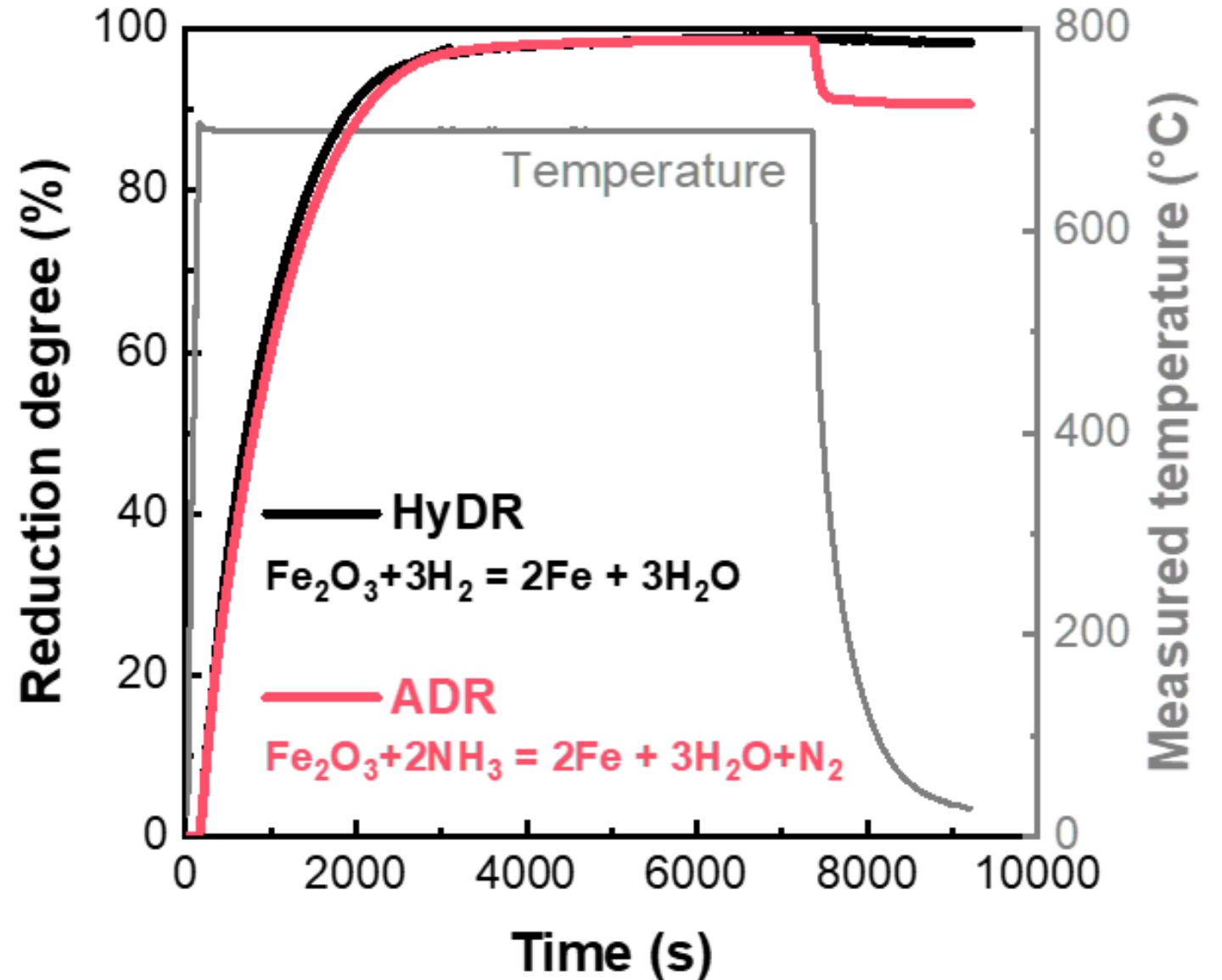


Source: Wood Mackenzie

Green hydrogen from Green Ammonia

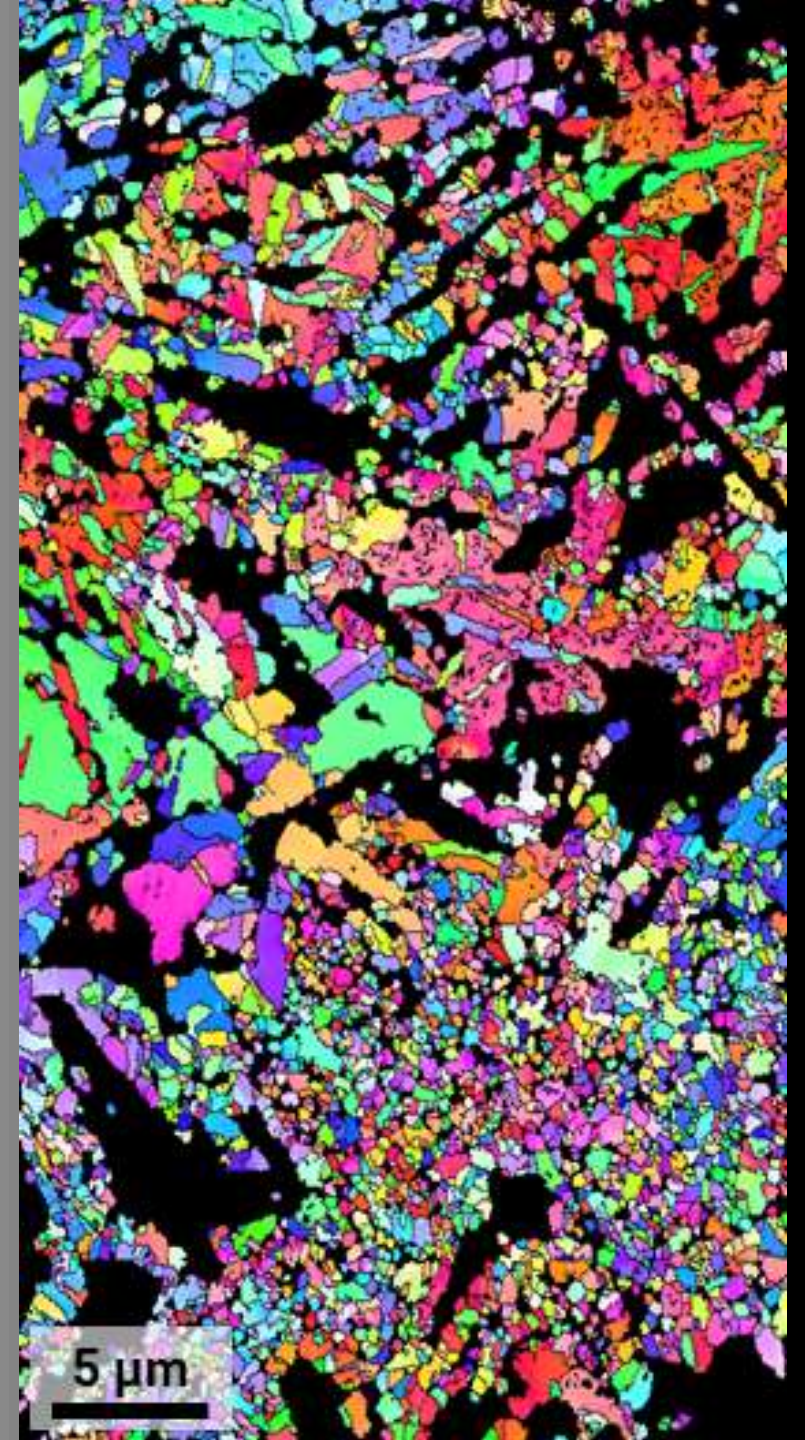
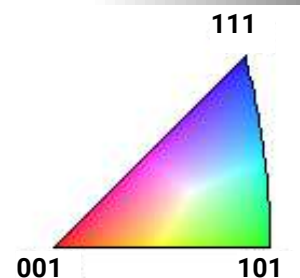


What is the kinetics?



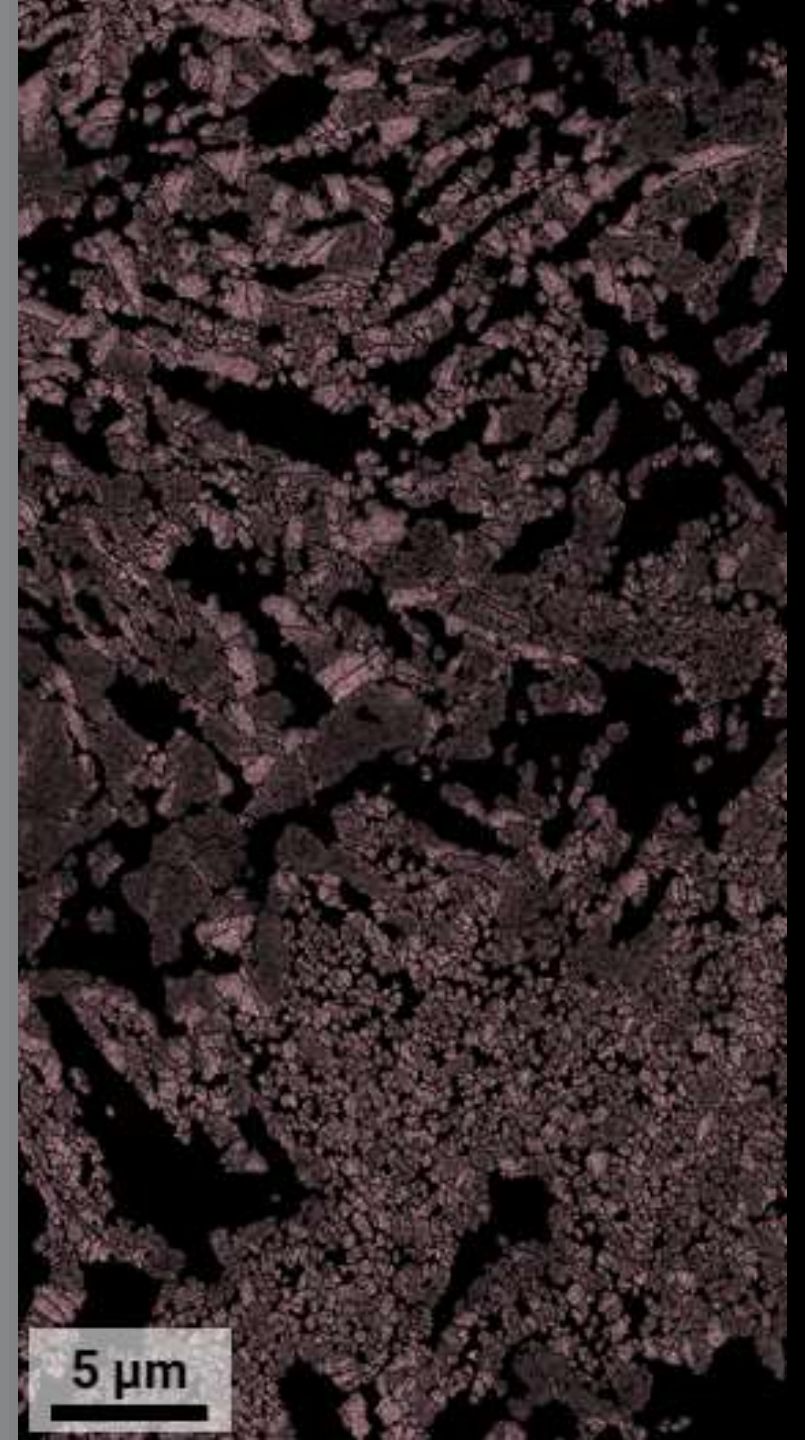
What kind of steel do we get from ammonia-based direct reduction

Ma, Y.; Bae, J. W.; Kim, S.; Jovič, M.; Li, K.; Vogel, D.; Ponge, D.; Rohwerder, M.; Gault, B.; Raabe, D. Reducing Iron Oxide with Ammonia: A Sustainable Path to Green Steel. *Adv. Sci.* 2023, 2300111, 1–7



Nitrogen in the steel

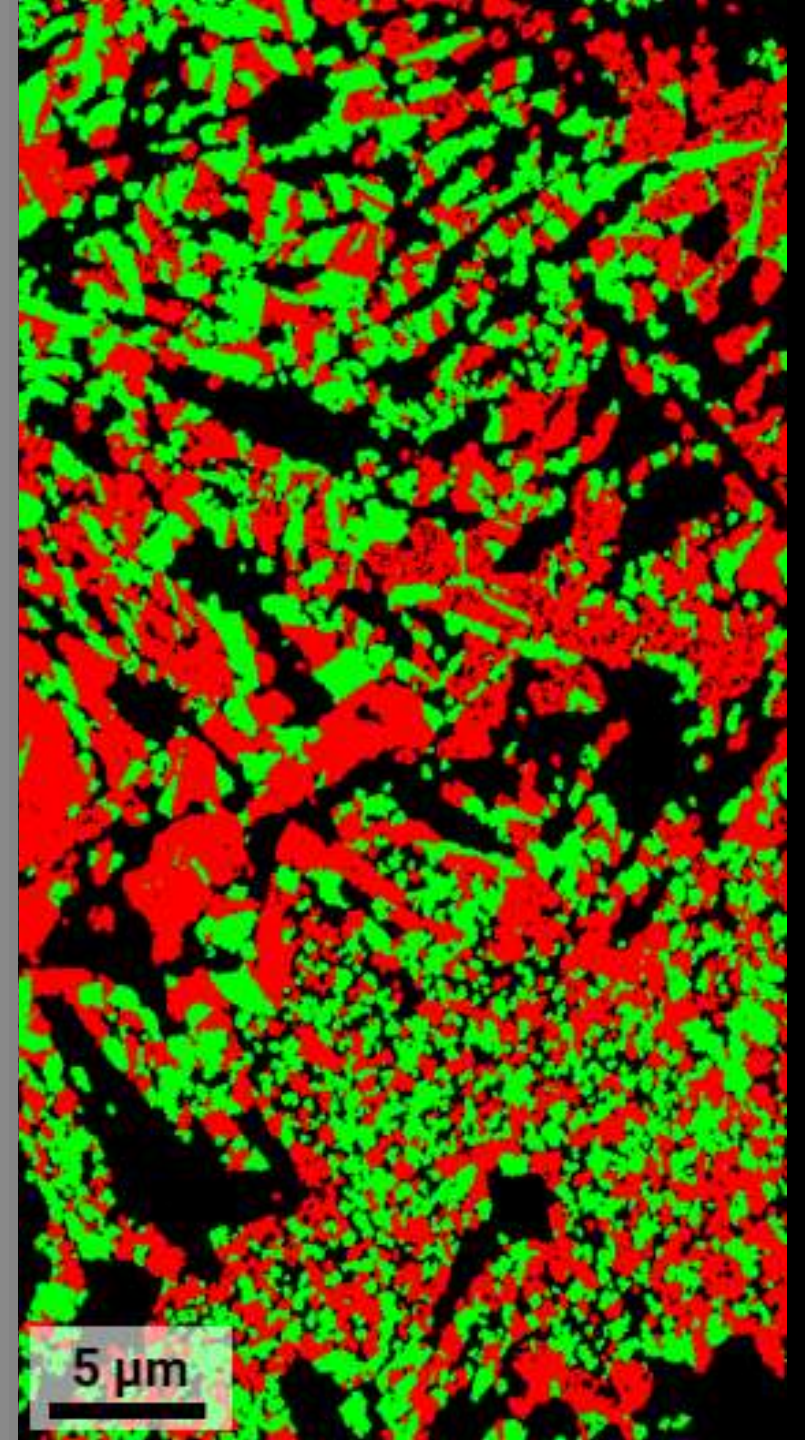
Ma, Y.; Bae, J. W.; Kim, S.; Jovič, M.; Li, K.; Vogel, D.; Ponge, D.; Rohwerder, M.; Gault, B.; Raabe, D. Reducing Iron Oxide with Ammonia: A Sustainable Path to Green Steel. *Adv. Sci.* 2023, 2300111, 1–7



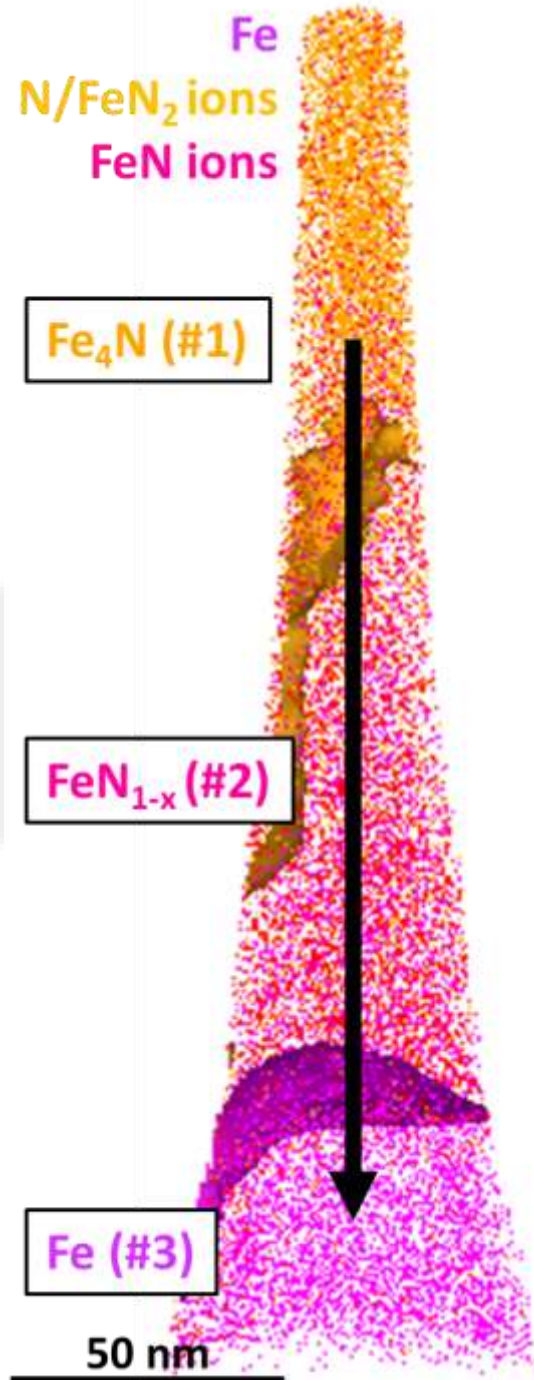
Nitrides in the steel

Ma, Y.; Bae, J. W.; Kim, S.; Jovič, M.; Li, K.; Vogel, D.; Ponge, D.; Rohwerder, M.; Gault, B.; Raabe, D. Reducing Iron Oxide with Ammonia: A Sustainable Path to Green Steel. *Adv. Sci.* 2023, 2300111, 1–7

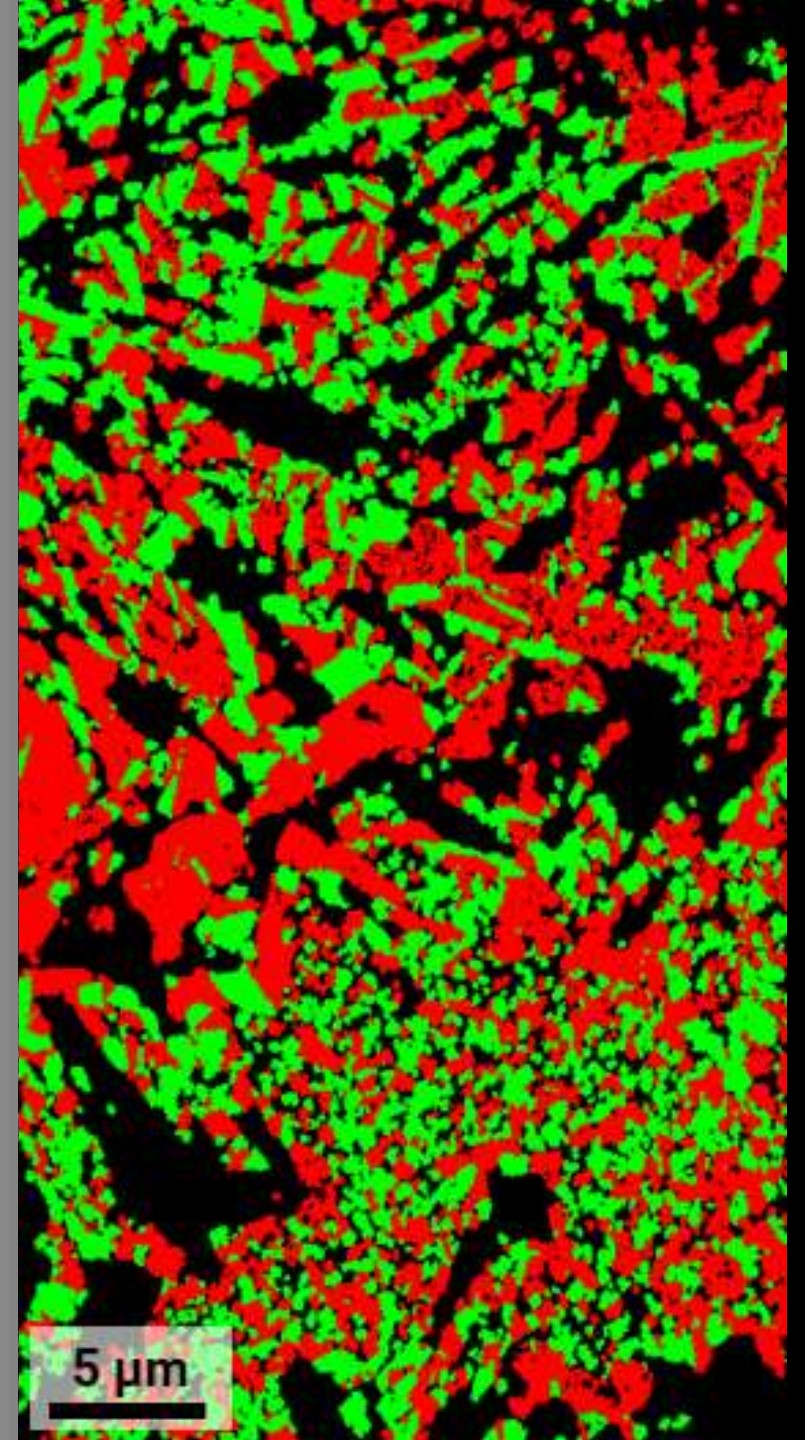
■ α -Fe
■ Fe_4N



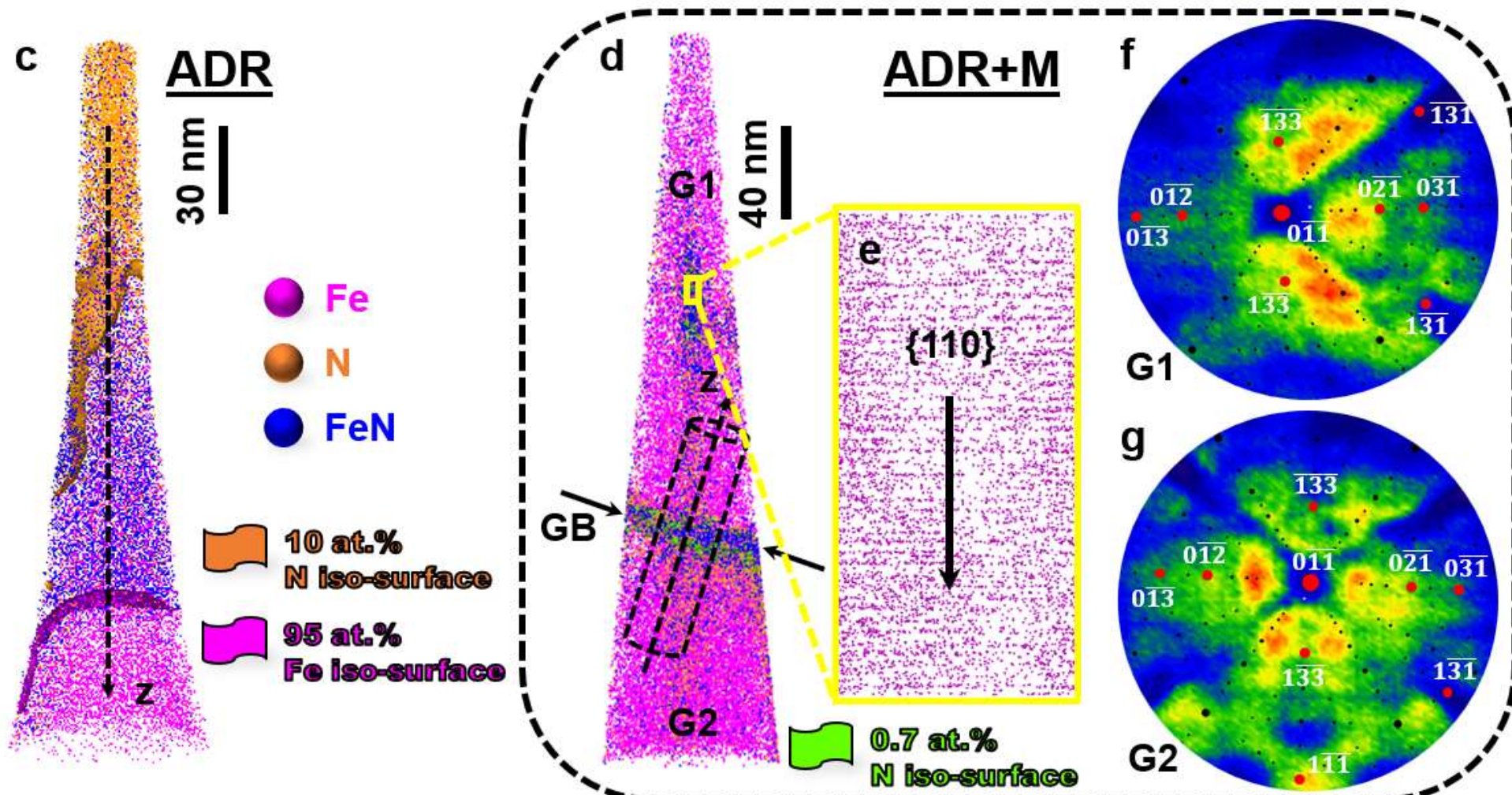
Nitrides in the steel



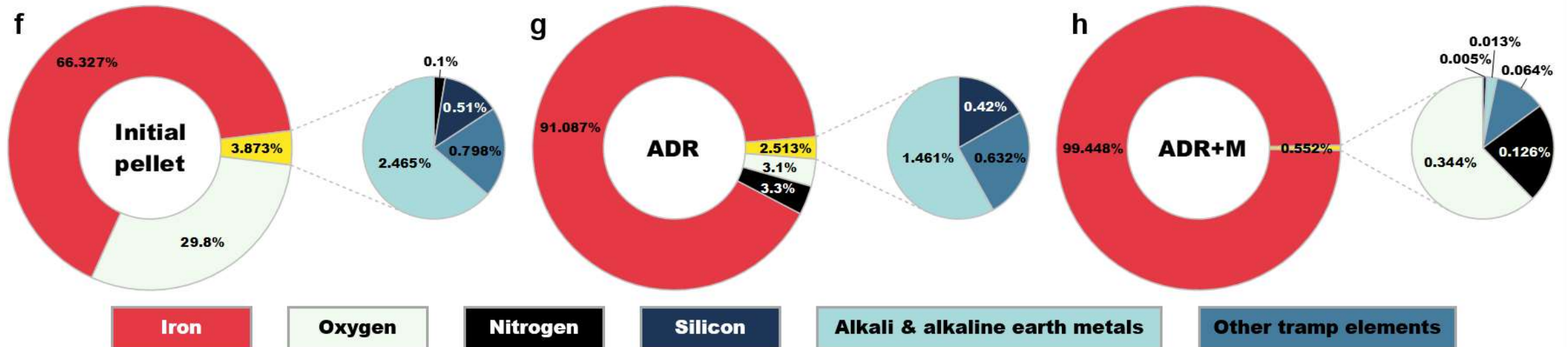
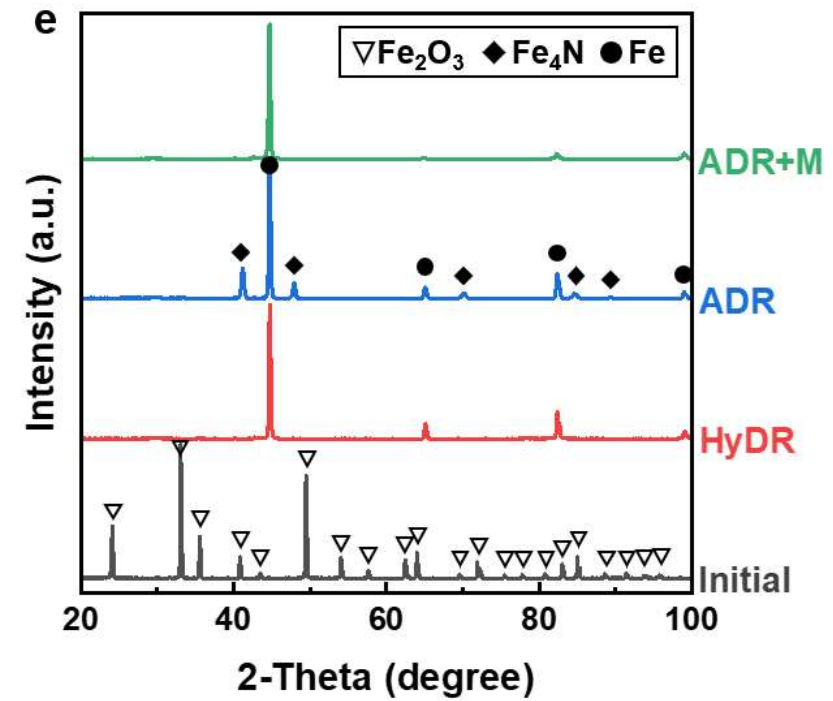
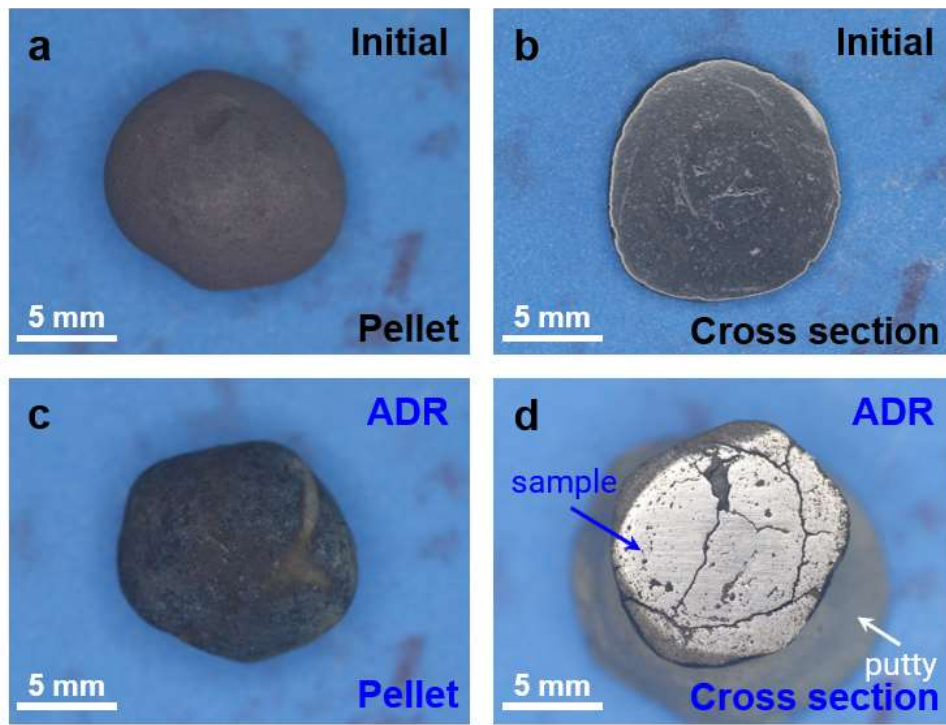
■ α -Fe
■ Fe₄N

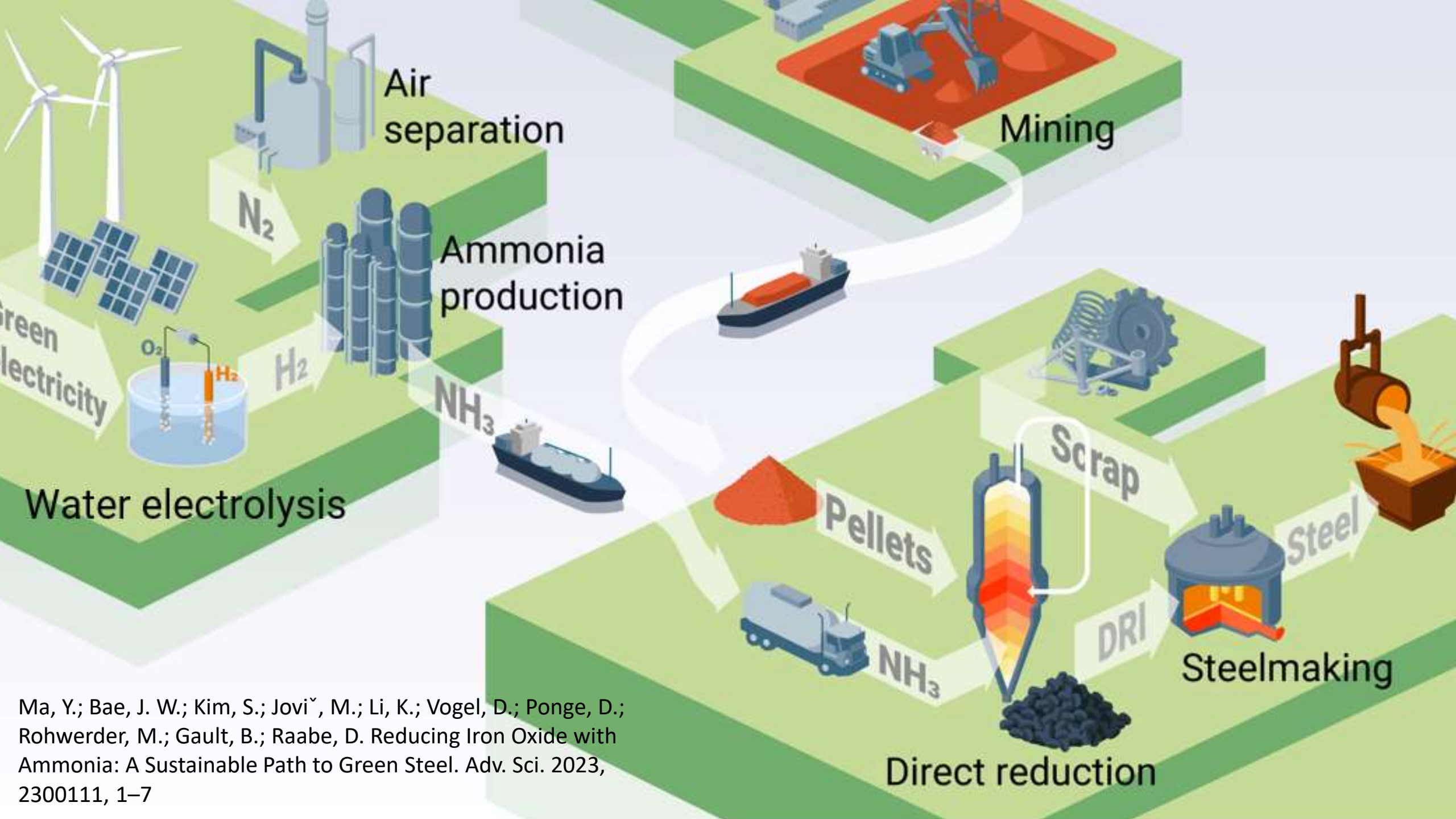


Nitrides in the steel

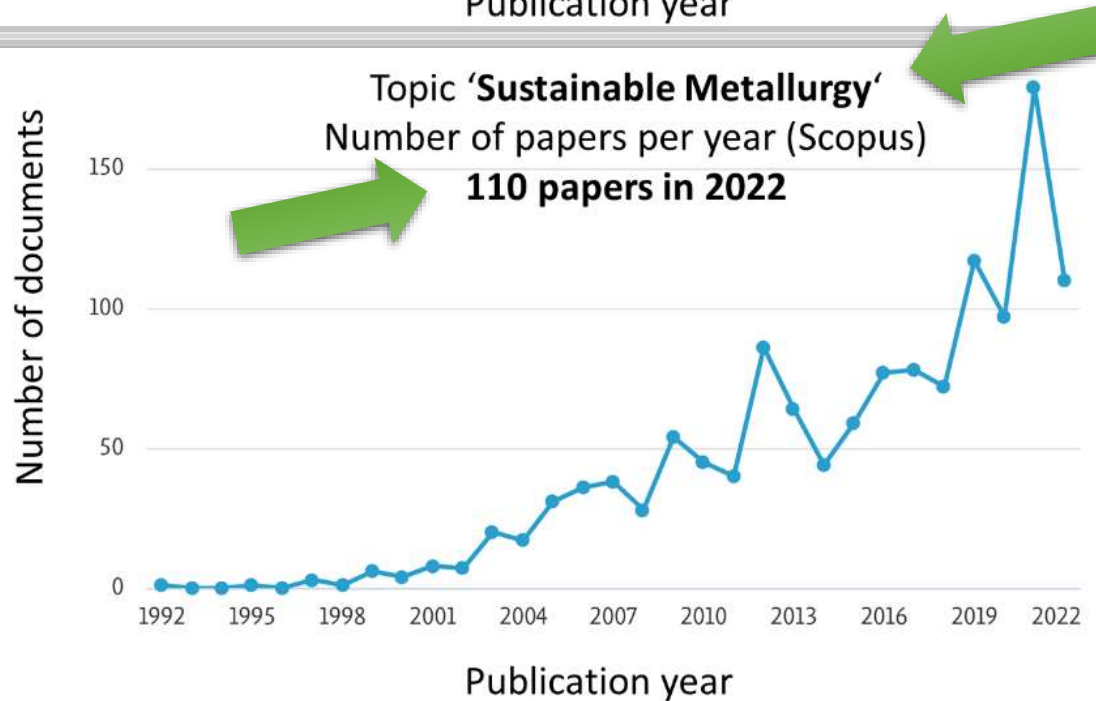
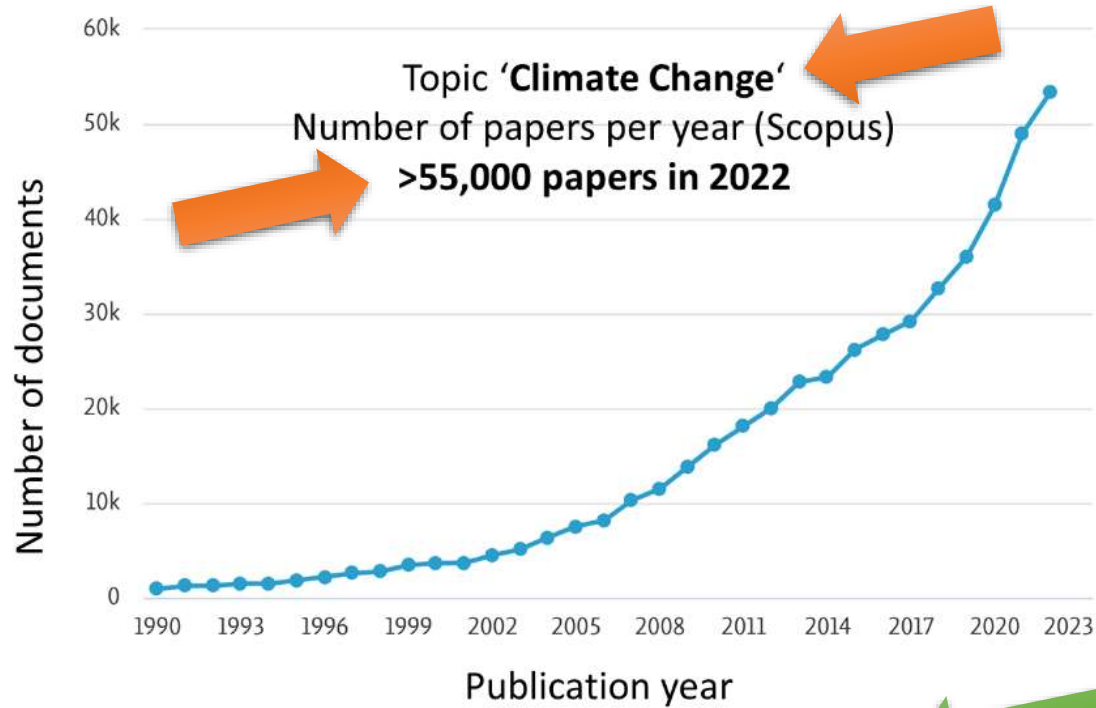


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Lots of headroom
for research that
helps to

REDUCE CO₂



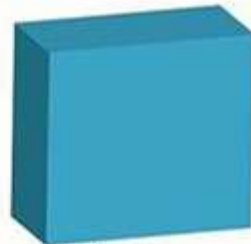
Decide !

Which one are you?

BE BOLD. BE THE CHANGE.



Critic

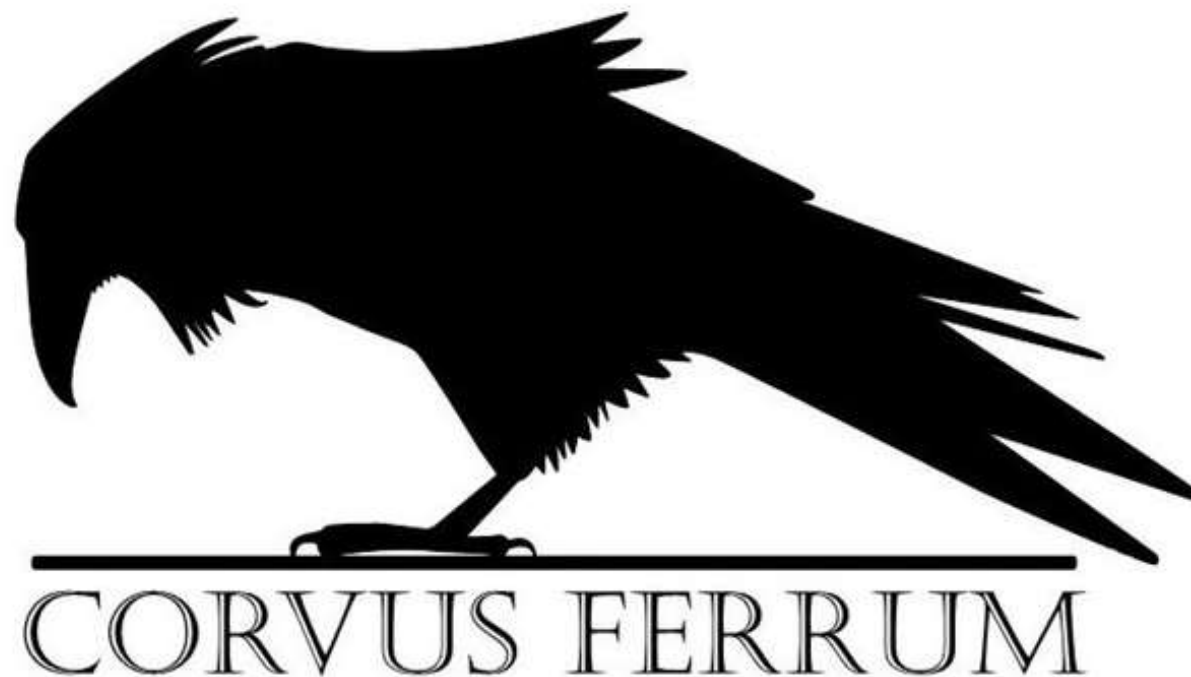


Talker



Doer





D. Raabe, The Materials Science behind Sustainable Metals and Alloys, Chem. Rev. 2023, 123, p. 2436

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<https://pubs.acs.org/doi/full/10.1021/acs.chemrev.2c00799>

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<https://www.nature.com/articles/s41586-019-1702-5>